New Sphingolipids from Marine Sponge *Iotrochota baculifera*

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Chemical examination of the marine sponge *Iotrochota baculifera* of the Indian Ocean furnished a sphingolipid (1), a new glycosphingolipid (iotroridoside-B, 2) and a four-component mixture of sphingolipids (3a—d) with two new components. Structure elucidation was carried out by spectral and chemical methods.

Key words *Iotrochota baculifera*; sphingolipids; iotroridoside-B; marine sponge; Myxillidae

Sphingolipids which are present widely in the membranes of eukaryotic cells of both terrestrial and marine organisms involve in many a physiological process and a few exhibited biological activities like cytotoxic, anti-inflammatory *etc.* as well.^{2–4)} We report in this communication the isolation and characterization of a sphingolipid (1), a glycosphingolipid (2) and a mixture of sphingolipid homologues (3a—d) from the marine sponge *Iotrochota baculifera* (Fam: Myxillidae) of the Indian Ocean. Earlier records on this genus include alkaloids, a toxin, ecdysteroids, sphingolipids, bromoindole and tyrosine derivatives.^{3,4)}

Results and Discussion

Separation of EtOAc extract by medium pressure liquid chromatograph (MPLC) yielded compounds 1, 2 and 3a—d as colourless crystals and showing the presence of hydroxyl and amide NH (3500—3200 cm⁻¹) and amide carbonyl (1640 cm⁻¹) in the IR spectra.

Compound 1, mp 145 °C, was analyzed for C₄₃H₈₇NO₅ by the fast atom bombardment mass spectrometry (FAB-MS) and elemental analysis. The ¹³C-NMR spectrum has signals appeared due to amide carbonyl at δ 175.3 and a methine carbon at δ 53.0. The ¹H-NMR spectrum showed signals of an amide proton doublet at δ 8.58 (1H, *J*=8.7 Hz) and protons of a long methylene chain at δ 1.29 (br s) indicating a sphingolipid skeleton.^{5,6)} The absence of singlets at δ 1.90—2.25 in ¹H-NMR spectra suggesting the absence of acetoxy groups in 1. Due to overlapping of signals in NMR spectra, 1 was acetylated and spectra recorded for acetate 1a. The ¹H- and ¹³C-NMR spectra showed four acetoxy methyl signals at δ 2.03—2.18 (12H, s) and 20.6—21.0, which revealed the presence of four hydroxyls in 1. The absence of a proton triplet at δ 2.22—2.50 in ¹H-NMR spectrum suggested the presence of —CO—CH(OAc)—CH₂— moiety in 1a.^{6,7)} The positions of oxygenated carbons in 1a were assigned on the basis of ¹H—¹H correlation spectroscopy (COSY) and heteronuclear multiple quantum coherence (HMQC) spectra which led to the proposal that 1 is a sphingolipid containing 1,3,4-trihydroxy base and a 2-hydroxy fatty acid and without unsaturation in fatty chains.

The length of the fatty acid and sphingosine chains in 1 were determined by methanolysis of 1 followed by acetylation of sphingamines according to method of Gaver and Sweeley.⁸⁾ The methyl ester and tetraacetylated sphingamines thus obtained were independently analysed by gas chromatography-mass spectrometry (GC-MS), which showed

ions at *m/z* 412 and 485 corresponding to methyl 2-hydroxy pentacosanoate and 1,3,4-triacetyl-2-acetamino-octadecane respectively. Based on the NMR and optical rotation values of natural and synthetic methyl esters and sphingamines,^{5,9)} *R* and 2*S*,3*S*,4*R* configurations were proposed to methyl ester and sphingamine moieties respectively. The structure of 1 could be described as (2*S*,3*S*,4*R*)-2-[(2*R*)-2-hydroxy pentacosanoylamino]-octadeca-1,3,4-triol. Literature has two reports of occurrence of this sphingolipid from non-marine sources, mushrooms *Phellinus ribis*¹⁰⁾ and *Grifola frondosa*.¹¹⁾

Compound 2, mp 205 °C, was analyzed for C₄₈H₉₃NO₁₀ from the positive FAB-MS and elemental analysis. The presence of an amide NH at δ 8.61 (d, *J*=9 Hz), a long chain of methylene protons at δ 1.29 (br s), two terminal methyls at δ 0.85 (t, *J*=6.9 Hz), an anomeric proton at δ 4.94 (d, *J*=8 Hz) and protons on oxygenated carbons appearing as multiplets between δ 3.86 and 4.75 in ¹H-NMR of 2 suggested it to be a

1, R=H; 1a, R=Ac

Fig. 1

2, R=H; 2a, R=Ac

Fig. 2

3a, n=14; 3b, n=15; 3c, n=16; 3d, n=17

Fig. 3

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glycosphingolipid.⁵⁾ The appearance of carbon resonances at δ 62.6, 71.5, 75.2, 78.5, 78.6 and 105.6 and a correlation between δ 4.94 and 105.6 in HMQC spectrum are attributable to β -glucopyranose as sugar moiety in **2**.^{5,12)}

The presence of seven acetoxy methyl groups resonating between δ 1.99 and 2.24 in ^1H -NMR spectrum of peracetylated derivative of **2** indicates the presence of three hydroxyls in the sphingolipid skeleton, in addition to four on glucopyranose unit of **2**. The positions of hydroxyl groups in the ceramide part were ascertained by ^1H - ^1H COSY data which assigned two double doublets at δ 4.75 and 4.55 and two multiplets at δ 4.34 and 4.20 to protons H-1a, H-1b, H-3 and H-4 respectively. The ^1H - ^1H COSY spectrum also showed the presence of a double bond and HMQC spectrum showed correlation between olefinic carbons at δ 132.4 and 126.3 with protons at δ 5.65 (1H, dt, $J=7.2$, 17.7 Hz) and 5.96 (1H, dt, $J=7.2$, 18.0 Hz) respectively. The olefinic proton at δ 5.96 showed correlation for two methylene protons each at δ 2.88 and 3.05. These two methylene protons are inter-correlated and correlated to a proton on the hydroxylated carbon appearing at δ 4.64 thus indicating the presence of $-\text{CO}-\text{CH}(\text{OH})-\text{CH}_2-\text{CH}=\text{CH}-$ moiety which was further evidenced by the absence of a triplet at δ 2.25–2.50.^{6,7)} From the observations it may be summarised that a methylene intervenes between the hydroxylated carbon and the olefinic carbon in fatty acyl chain. The other olefinic proton at δ 5.65 (H-5') showed cross peaks with methylene protons at δ 2.16, which further showed cross peaks with long-chain methylene protons at δ 1.29. The coupling constant between the olefinic protons (17.7 Hz) and chemical shifts of the adjacent carbons at δ 33.6 and 32.2 strongly suggest *trans* (*E*) orientation (a cisoid orientation is expected to have a J value of 6–12 Hz and adjacent carbon shifts at ≈ 27 ppm).^{5,13,14)} All the assignments were further confirmed by HMQC spectra.

The fatty acid and sphingosine chain lengths and sugar of **2** were determined through methanolysis according to Gaver and Sweeley.⁸⁾ ^1H -NMR of the ester so obtained on methanolysis showed the presence of olefinic protons at δ 5.34 (1H, dt, $J=17.7$, 7.2 Hz, H-5) and 5.54 (1H, dt, $J=7.2$, 17.4 Hz, H-4). GC-MS analysis showed the presence of the methyl 2-hydroxy 4-tetracosenoate (m/z 396). On comparison with the literature data,⁵⁾ the optical rotation $[\alpha]_D^{28} -2.4^\circ$ indicated *R* configuration for the ester. The sphingosine base and methyl β -glucoside were separated from aqueous methanolic layer. The sphingosine base, after acetylation followed by chromatography over Sephadex LH20, gave the tetraacetyl sphingamine moiety which on GC-MS analysis showed one ion at m/z 485 for $[\text{M}^+]$ and m/z 426 for $[\text{M}-\text{AcO}]^+$, corresponding to the 1,3,4-triacetoxy-acetamino octadecane. On the basis of NMR and optical rotation data in the literature,^{5,9)} 2*S*, 3*S*, 4*R* configuration could be proposed for the sphingamine base. The sugar residue obtained from the aqueous layer was identified as methyl- β -D-glucopyranoside by comparing the spectral data and optical rotation with those reported for authentic sample.⁵⁾

From the foregoing discussion, compound **2** could be characterized as (2*S*,3*S*,4*R*)-1-*O*-(β -D-glucopyranosyl)-2-[(2*R*,4*E*)-2-hydroxy-4-tetracosenoylamino]-octadecane-1,3,4-triol. As literature survey indicated that the compound iotroridoside-A¹⁵⁾ has the same structure excepting '*Z*' configuration at the double bond, **2** was named as iotroridoside-B.

The FAB-MS of **3** showed a series of pseudomolecular ion peaks $[\text{M}+\text{H}]^+$ at m/z 654, 668, 682 and 696 suggesting a mixture of homologues. As attempts to separate the constituents by HPLC including reverse phase were not successful, the spectra were recorded for the mixture. A few differences were noticed by a scrutiny of spectral data of **1** and **3** viz. the presence of a triplet at δ 2.45 and three acetoxy methyls in ^1H -NMR spectrum of **3** and its acetate respectively, showing three hydroxyls and a non-hydroxy fatty acid in **3**. The positions of hydroxyls in **3** were assigned on the basis of ^1H - ^1H COSY data (see experimental section).

The length of the fatty acid and sphingosine chains in **3** were determined by methanolysis⁸⁾ which gave methyl ester and sphingamines. The GC-MS of the methyl ester showed an ion at m/z 354 corresponding to methyl docosanoate whereas the acetylated sphingamines showed four ions at m/z 499 (23%), 513 (26%), 527 (23%) and 541 (28%) corresponding to 1,3,4-triacetyl-2-acetamino-substituted nonadecane, eicosane, heneicosane and docosane respectively. From the foregoing discussion it may be proposed that **3** is a mixture of sphingolipids containing 1,3,4-trihydroxy-2-docosanoylamino-nonadecane (**3a**), 1,3,4-trihydroxy-2-docosanoylamino-eicosane (**3b**), 1,3,4-trihydroxy-2-docosanoylamino-heneicosane (**3c**) and 1,3,4-trihydroxy-2-docosanoylamino-docosane (**3d**). Of this **3a** and **3c** are new to the literature while **3b** and **3d** have been reported earlier.^{16,17)}

Experimental

General ^1H - and ^{13}C -NMR spectra were recorded on a Bruker DRX spectrometer operating at 300 MHz and 75 MHz respectively. The chemical shift values were reported in parts per million units and the coupling constants were in Hz. FAB-MS was recorded on a JEOL-Sx-120/DA-6000 mass spectrometer using a beam of Argon/Xenon (2–8 keV) and *m*-nitrobenzyl alcohol as the matrix. Optical rotations were taken on a JASCO DIP-370 polarimeter. Elemental analysis was carried out on a Carlo Erba 1108 analyzer. IR spectra were recorded on a Perkin-Elmer 881 instrument. Melting points were recorded on Boitus melting point apparatus and were uncorrected. Si gel column chromatography was carried out using silica gel (finer than 200#, ACME), gel filtration was carried out using LH20 (Sephadex LH20, Pharmacia Biotech) and MPLC was performed on Buchi B-688 MPLC system. GC-MS experiments were carried out on Shimadzu's GCMS-QP 5050A system. HPLC was carried out on Shimadzu HPLC class VP series with two LC-10AT VP pumps and UV/Vis Detector SPD-10A VP.

Animal Material Specimens of the sponge *Iotrochota baculifera* were collected from the Coast near Tuticorin, India ($8^\circ 45'\text{N}$, $78^\circ 12'\text{E}$) at a depth of 10–15 m and preserved in methanol until extraction. A voucher specimen was deposited in the Marine organisms museum in the Department of Pharmaceutical Sciences, Andhra University, Visakhapatnam, India (Voucher No. AU2-152).

Extraction and Isolation The organism was extracted with MeOH eight times at room temperature on every 48 h. The combined alcoholic extract was concentrated under reduced pressure. The concentrated extract was fractionated with EtOAc and the EtOAc fractions were pooled and concentrated under reduced pressure after drying on anhydrous MgSO_4 to yield a crude residue (21 g). The residue (18 g) was column chromatographed over silica gel (400 g, finer than 200#, ACME, column diameter 60 mm) using eluants with increasing polarity starting from *n*-hexane through EtOAc to MeOH. The fractions eluted with 40% and 60% EtOAc in hexane and EtOAc were pooled and concentrated to yield **3a–d** (40 mg), **1** (120 mg) and **2** (28 mg) respectively. The mixture **3a–d** was inseparable by reverse phase HPLC on C_{18} column (MeOH).

Compound 1: $[\alpha]_D^{28} +10.2^\circ$ ($c=0.45$, pyridine). IR (KBr) cm^{-1} : 3460, 1630. ^1H -NMR (pyridine- d_5 , 300 MHz) δ : 8.58 (1H, d, $J=8.7$ Hz, NH), 5.09 (1H, m, H-2), 0.87 (6H, t, $J=6.8$ Hz, CH_3). ^{13}C -NMR (pyridine- d_5 , 75 MHz) δ : 62.1 (t, C-1), 175.3 (s, C-1'), 53.0 (d, C-2), 73.0 (d, C-2'), 76.8 (d, C-3), 34.8 (t, C-3'), 72.5 (d, C-4), 34.2 (t, C-5), 30.4–25.9 (t), 14.3 (q, end methyls). FAB-MS m/z : 698 $[\text{M}+\text{H}]^+$, 382, 300, 282, 264. *Anal.* Calcd for $\text{C}_{43}\text{H}_{87}\text{NO}_5$: C, 73.96; H, 12.57; N, 2.01. Found: C, 74.09; H, 12.61; N, 1.97.

Compound **2**: $[\alpha]_D^{28} + 12.0^\circ$ ($c=0.15$, pyridine). IR (KBr) cm^{-1} : 3500—3200, 2940, 1630. $^1\text{H-NMR}$ (pyridine- d_5 , 300 MHz) δ : 8.61 (1H, d, $J=9$ Hz, NH), 4.75 (1H, dd, $J=7$, 10 Hz, H-1a), 4.55 (1H, dd, $J=4$, 10 Hz, H-1b), 5.29 (1H, m, H-2), 4.34 (1H, dd, $J=5$, 7 Hz, H-3), 4.20 (1H, m, H-4), 1.90 (2H, m, H-5), 1.66 (2H, m, H-6), 4.64 (1H, dd, $J=8$, 4 Hz, H-2'), 2.88 (1H, m, H-3a'), 3.05 (1H, m, H-3b'), 5.96 (1H, dt, $J=7.2$, 18.0 Hz, H-4'), 5.65 (1H, dt, $J=7.2$, 17.7 Hz, H-5'), 2.16 (2H, m, H-6'), 1.29 (56H, brs, $-\text{CH}_2-$), 0.85 (6H, t, $J=6.9$ Hz, end methyls), 4.94 (1H, d, $J=8$ Hz, H-1''), 4.03 (1H, dd, $J=8$, 9 Hz, H-2''), 4.18 (1H, m, H-3''), 4.21 (1H, m, H-4''), 3.86 (1H, brs, H-5''), 4.31 (1H, dd, $J=6$, 12 Hz, H-6a''), 4.50 (1H, dd, $J=4$, 12 Hz, H-6b''). $^{13}\text{C-NMR}$ (pyridine- d_5 , 75 MHz) δ : 70.5 (t, C-1), 51.8 (d, C-2), 75.8 (d, C-3), 72.5 (d, C-4, C-2'), 34.2 (t, C-5), 175.0 (s, C-1'), 33.6 (t, C-3'), 126.3 (d, C-4'), 132.4 (d, C-5'), 32.2 (t, C-6'), 30.4—23.0 (t, $-\text{CH}_2-$), 14.3 (q, end methyls), 105.6 (d, C-1''), 75.2 (d, C-2''), 78.5 (d, C-3''), 71.5 (d, C-4''), 78.6 (d, C-5''), 62.6 (t, C-6''). FAB-MS m/z : 844 [M+H], 664, 348, 300, 282, 264. *Anal.* Calcd for $\text{C}_{48}\text{H}_{93}\text{NO}_{10}$: C, 68.27; H, 11.11; N, 1.66. Found: C, 68.19; H, 11.24; N, 1.73.

Compound **3a—d**: Mixture melted at 121—123 °C. IR (KBr) cm^{-1} : 3500, 1640. $^1\text{H-NMR}$ (pyridine- d_5 , 300 MHz) δ : 8.50 (d, $J=7.4$ Hz, NH), 4.49 (dd, $J=3$, 11 Hz, H-1a), 4.48 (dd, $J=4.2$, 11 Hz, H-1b), 5.08 (m, H-2), 4.39 (dd, $J=5$, 3 Hz, H-3), 4.28 (m, H-4), 1.96 (m, H-5), 1.69 (m, H-6), 2.45 (t, $J=7.5$ Hz, H-2'), 2.22 (m, H-3'), 1.85 (m, H-4'), 1.30 (brs, $-\text{CH}_2-$), 0.86 (t, $J=3.2$ Hz, end methyls). $^{13}\text{C-NMR}$ (pyridine- d_5 , 75 MHz) δ : 62.2 (t, C-1), 53.8 (d, C-2), 76.7 (d, C-3), 73.1 (d, C-4), 33.9 (t, C-5), 173.4 (s, C-1'), 36.9 (t, C-2'), 32.1 (t, C-3'), 32.1 (t, C-4'), 30.3—23.0 (t, $-\text{CH}_2-$), 14.3 (q, end methyls). FAB-MS m/z : 654, 668, 682, 696 [M+H]⁺.

Acetylation Dry pyridine (0.5 ml) and Ac_2O (1.0 ml) were added to compounds **1** (25 mg), **2** (10 mg) and **3** (10 mg) separately and left overnight. After usual workup and crystallisation in each case yielded **1a**, **2a** and **3e**.

Compound **1a**: mp 61 °C, $[\alpha]_D^{28} + 11.0^\circ$ ($c=0.1$, CHCl_3). $^1\text{H-NMR}$ (CDCl_3 , 300 MHz) δ : 6.62 (1H, d, $J=9$ Hz, NH), 4.34 (1H, dd, $J=6.6$, 11.4 Hz, H-1a), 4.02 (1H, dd, $J=3$, 11.7 Hz, H-1b), 4.44 (1H, m, H-2), 5.10 (2H, dd, $J=6.9$, 10.8 Hz, H-3, H-2'), 4.95 (1H, m, H-4), 1.69 (4H, m, H-5, H-4'), 1.84 (2H, m, H-3'), 1.25 (64H, brs, $-\text{CH}_2-$), 0.85 (6H, t, $J=6.0$ Hz, end methyls), 2.03—2.18 (12H, s, 4-COCH₃). $^{13}\text{C-NMR}$ (CDCl_3 , 75 MHz) δ : 62.4 (t, C-1), 47.8 (d, C-2), 74.0 (d, C-3), 72.3 (d, C-4), 31.8 (t, C-5), 171.3 (s, C-1'), 72.7 (d, C-2'), 31.9 (t, C-3'), 31.8 (t, C-4'), 29.7—22.7 (t, $-\text{CH}_2-$), 14.1 (q, $-\text{CH}_3$), 20.6—21.0 (s, $-\text{COCH}_3$), 170.0—170.9 (s, COCH_3). FAB-MS m/z : 866 [M+H]⁺.

Compound **2a**: mp 57 °C. $^1\text{H-NMR}$ (CDCl_3 , 300 MHz) δ : 6.80 (1H, d, $J=9.0$ Hz, NH), 4.28 (1H, m, H-2), 1.99—2.24 (21H, s, 7-COCH₃), 1.25 (brs, $-\text{CH}_2-$), 0.86 (6H, t, $J=6.8$ Hz, end methyls).

Compound **3e**: mp 65 °C. $^1\text{H-NMR}$ (CDCl_3 , 300 MHz) δ : 6.62 (d, $J=9$ Hz, NH), 1.90—2.01 (s, $-\text{COCH}_3$), 1.25 (brs, $-\text{CH}_2-$), 0.85 (t, $J=5.7$ Hz, end methyls). FAB-MS m/z : 766, 780, 794, 808 [M+H]⁺.

Methanolysis Compounds **1** and **3** (10 mg) were separately treated with 3 ml of 1 N HCl in methanol at 90 °C for 15 h with magnetic stirring. The fatty acid methyl ester so obtained in each case was extracted with *n*-hexane and analyzed by GC-MS. In each case a single compound was obtained. Methyl ester from Compound **1**: $[\alpha]_D^{28} - 2.6^\circ$ ($c=0.05$, CHCl_3). $^1\text{H-NMR}$ (CDCl_3 , 300 MHz) δ : 5.13 (1H, dd, $J=6.9$, 3.9 Hz, H-2), 3.52 (3H, s, OCH₃), 1.84 (2H, m, H-3), 1.68 (2H, m, H-4), 1.25 (42H, brs, $-\text{CH}_2-$), 0.88 (3H, t, CH₃). GC-MS m/z : 412 [M]⁺. Methyl ester from Compound **3**: $^1\text{H-NMR}$ (CDCl_3 , 300 MHz) δ : 2.28 (2H, t, $J=7.2$ Hz, H-2), 3.57 (3H, s, OCH₃), 1.84 (2H, m, H-3), 1.30 (36H, brs, $-\text{CH}_2-$), 0.85 (3H, t, $J=7.5$ Hz, CH₃). GC-MS m/z : 354 [M]⁺. The MeOH/H₂O phase was evaporated and the residue was acetylated. Purification by filtration over a Sephadex LH20 column ($\text{CH}_2\text{Cl}_2/\text{MeOH}$, 1:1) gave the acetylated sphingamines. The sphingamines were subjected to GC-MS in each case. Acetylated sphingamines from Compound **1**: $[\alpha]_D^{28} + 24.2^\circ$ ($c=0.05$, CHCl_3). $^1\text{H-NMR}$ (CDCl_3 , 300 MHz) δ : 6.61 (1H, d, $J=9$ Hz, NH), 5.10 (1H, dd, $J=7$, 4 Hz, H-3), 4.96 (1H, m, H-4), 4.44 (1H, m, H-2), 4.35 (1H, dd, $J=6.2$, 10.4 Hz, H-1a), 4.03 (1H, dd, $J=4.1$, 10.6 Hz, H-1b), 2.18—2.03 (12H, s, 4-COCH₃), 1.68 (2H, m, H-5), 1.25 (24H, brs, $-\text{CH}_2-$), 0.88 (3H, t, $J=5.8$ Hz, CH₃). GC-MS m/z : 485 [M]⁺. Acetylated sphingamines from Compound **3a—d**:

$^1\text{H-NMR}$ (CDCl_3 , 300 MHz) δ : 6.60 (d, $J=9$ Hz, NH), 5.10 (dd, $J=4.5$, 7.8 Hz, H-3), 4.93 (dd, $J=3.7$, 9.8 Hz, H-1b), 4.63 (m, H-4), 4.44 (m, H-2), 4.39 (dd, $J=4.6$, 9.6 Hz, H-1a), 2.18—2.03 (s, 4-COCH₃), 1.84 (m, H-4), 1.65 (m, H-5), 1.25 (brs, $-\text{CH}_2-$), 0.85 (t, $J=6.2$ Hz, CH₃). GC-MS m/z : 499 (23%), 513 (26%), 527 (23%), 541 (28%) [M]⁺.

The process was repeated with compound **2** (8 mg). The methyl ester so obtained was analysed by GC-MS which showed single component with m/z : 396 [M]⁺. $[\alpha]_D^{28} - 2.4^\circ$ ($c=0.05$, CHCl_3). $^1\text{H-NMR}$ (CDCl_3 , 300 MHz) δ : 5.54 (1H, dt, $J=17.4$, 7.2 Hz, H-4), 5.34 (1H, dt, $J=17.7$, 7.2 Hz, H-5), 4.73 (1H, dd, $J=8$, 4 Hz, H-2), 3.61 (3H, s, OCH₃), 2.18 (1H, m, H-3a), 1.98 (1H, m, H-3b), 1.25 (34H, brs, $-\text{CH}_2-$), 0.86 (3H, t, $J=8.4$ Hz, end methyl). The aqueous methanolic layer of the hydrolysate was evaporated to dryness and the residue was separated by TLC as sphingosine base and methylated sugar. The base was acetylated and subjected to GC-MS m/z : 485 [M]⁺, $[\alpha]_D^{28} + 26.7^\circ$ ($c=0.05$, CHCl_3). $^1\text{H-NMR}$ (CDCl_3 , 300 MHz) δ : 6.81 (1H, d, $J=9$ Hz, NH), 5.30 (1H, dd, $J=4.1$, 7.3 Hz, H-3), 4.95 (1H, m, H-4), 4.43 (1H, dd, $J=10.2$, 3.9 Hz, H-1b), 4.28 (1H, m, H-2), 4.12 (1H, dd, $J=10.2$, 4.3 Hz, H-1a), 1.78 (2H, m, H-5), 1.56 (2H, m, H-6), 2.02—2.12 (12H, s, 4-COCH₃), 1.25 (22H, brs, $-\text{CH}_2-$), 0.86 (3H, t, $J=8.4$ Hz, end methyl). The presence of methyl- β -D-glucopyranose was established by comparison of NMR data with those reported in the literature and with an authentic sample $[\alpha]_D^{28} + 74^\circ$ ($c=0.1$, MeOH), R_f 0.46 (EtOAc/MeOH/H₂O, 5:2:0.5). Authentic sample $[\alpha]_D^{28} + 77^\circ$ ($c=0.1$, MeOH), R_f 0.46.

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References and Notes

- Present address: Bhargavi, 9-40-7, Pithapuram Colony, Visakhapatnam, India.
- Karlsson K. A., "Biological Membranes," Vol. 4, ed. by Chapman D., Academic Press, London, 1982, p. 1.
- John W. B., Brent R. C., Murray H. G. M., Peter T. N., Michèle R. P. *Nat. Prod. Rep.*, **20**, 1—48 (2003).
- Faulkner D. J., *Nat. Prod. Rep.*, **19**, 1—48 (2002), and earlier annual reports.
- Jin W., Rinehart K. L., Jares-Erijman E. A., *J. Org. Chem.*, **59**, 144—147 (1994).
- Chebaane K., Guyot M., *Tetrahedron Lett.*, **27**, 1495—1496 (1986).
- Bheemasankara Rao Ch., Satyanarayana Ch., *Indian J. Chem.*, **33B**, 97—98 (1994).
- Gaver R. C., Sweeley C. C., *J. Am. Oil. Chem. Soc.*, **42**, 294—298 (1965).
- Sugiyama S., Honda M., Komori T., *Liebigs Ann. Chem.*, **1988**, 619—625 (1988).
- Moon D.-C., Hwang K.-H., Choi K.-R., Lee Y.-M., Kim J.-H., Kim J.-G., Park Y.-H., Kim M.-R., Zee O.-P., *Anal. Sci. Technol.*, **7**, 547—554 (1994).
- Yaoita Y., Ishizuka T., Kakuda R., Machida K., Kikuchi M., *Chem. Pharm. Bull.*, **48**, 1356—1358 (2000).
- Bock K., Pederson C., "Advances in Carbohydrate Chemistry and Biochemistry," Vol. 41, eds. by Jipson R. S., Horfon D., Academic Press, New York, 1983, pp. 27—46.
- Silverstein R. M., Webster F. X., "Spectrometric Identification of Organic Compounds," 6th ed., John Wiley & Sons, Inc., New York, 1998.
- Fusetani N., Yasumoto K., Matsunaga S., Hirota H., *Tetrahedron Lett.*, **30**, 6891—6894 (1989).
- Deng S.-Z., Tian C.-L., Xiao D.-J., Wu H.-M., *Chin. J. Chem.*, **19**, 362—364 (2001).
- Meng Y., Su J., Zeng L., *Zhongshan Daxue Xuebao, Ziran Kexueban*, **35**, 69—72 (1996).
- Wild R., Schimdt R. R., *Liebigs Ann. Chem.*, **1995**, 755—764 (1995).