MDP-based universal adhesive system irradiated with Er,Cr:YSGG: Analysis of its performance up to 6 months

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This study aimed to analyze the interaction of a MDP-based universal dentin bonding system (DBS) with Er, Cr: YSGG laser irradiation, measuring the microtensile bond strength to dentin over a six-month period. The experimental design involved three factors: DBS (Adper Scotchbond Universal; Clearfil SE Bond, Adper Scotchbond Multipurpose and Adper Single Bond 2), laser (Control and Er, Cr: YSGG), and time (initial- 7 days and 6 months). Eighty dentin samples from molars were prepared (n=10) with laser irradiation after primer and DBS application. After 7 days, were subjected to micro tensile bond strength test. The data were analyzed by three-way ANOVA and Tukey tests (p<0.05). Both DBS and laser significantly affected the bonding performance and their interaction was statistically significant (p=0.0194). The self-etching mode of the MDP-based universal DBS maintained the bond strength on dentin irradiated with ER, Cr: YSGG after 6 months, while bonding with all other DBS deteriorated.

Keywords: Dentin-bonding agents, Laser, Micro tensile bond strength, Universal adhesive

INTRODUCTION

Bonding to dentin occurs via a complex mechanism as it involves dynamic biological tissues1-3. Therefore, different categories of dentin bonding systems were developed to overcome the main challenges regarding the establishment of a long-term interface between these polymeric materials and the biological tissues4-7.

To this end, understanding the compositional and structural formation as well as the mineral content and organic matrix of dentin is needed to guide the development of improved adhesive materials8-10. Novel materials that consider the interactions between polymers and the dentin under clinical conditions have potential to exhibit enhanced performance11,12.

Regarding dentin bonding systems (DBS), universal or multi-mode DBS have been used in many investigations13-15,11,12. The most attractive feature of these systems is the possibility to use them as part of etch-and-rinse or self-etching techniques. Some systems also provide the choice of dry or moist bonding technique protocols.

With previously reported evidence of good performance, the use of universal DBS in combination with other strategies to improve the interaction between substrate and MDP-based DBS has become increasingly popular16-18. Stable chemical bonding is also achieved19,20 and laser association is a promising tool that has already been used to enhance bonding to dentin19,21.

Initially, several investigations assessed the effect of the use of erbium lasers for dental cavity treatments19,20 to morphologically modify the enamel and dentin substrate through interactions with hard dental tissue21,22. Its use has been widely reported in the literature23-25. The erbium, chromium: yttrium scandium gallium garnet (Er, Cr: YSGG) laser irradiates at 2,790 nm, where the maximum absorption peak of hydroxyapatite and water occurs. This allows the laser to cut hard tissues such as bone, enamel, and dentin26. Irradiated on dentin can create a substrate free of the smear layer, denaturing the organic content, and reducing the solubility of hydroxyapatite, which is suggested to improve the adhesive interface27-29. Controversially, some studies have observed flaws in the morphological modifications of irradiated dentin, including opened dentinal tubules, fissures, and cracks on the peritubular dentin which appeared protruded on dentin surface29,30. The use of a laser with a wavelength compatible with the water absorption peak could be an interesting strategy for the removal of residual water present within the dentin structure responsible for the hydrolytic degradation that occurs over time, decreasing the clinical time of restoration31,32.

Previous studies assumed that laser irradiation could provide long lasting dental bonding performance if used during, instead of before, DBS application. In 1999 G onçalves et al.13 used irradiation with an experimental high-power laser (Nd: YFL) after adhesive application before light-curing and obtained promising results with increased shear bonding strength to dentin. Marimoto et al.14 used Nd: YAG (1,064 nm) high-power laser to compare the bond strength of two-step etch-and-rinse systems and self-etch systems irradiated under the same conditions. The authors also observed increased bonding strength when the adhesives were irradiated.10 Maenosono et al.15 proposed the use of a diode laser (970
nm), which is portable and inexpensive. In their study, the bond strength of the simplified dental adhesive was evaluated, showing increased bond strength for the laser irradiated groups. Based on these investigations, the use of lasers for etch-and-rinse DBS improved the formation of more uniform hybrid layers\(^\text{[17]}\). This is one of the most accepted hypotheses regarding the effective use of lasers with solvated etch-and-rinse systems\(^\text{[17]}\).

The use of adhesives containing functional monomers with the ability to promote chemical interaction with calcium ions has become the more advocated strategy recently\(^\text{[31]}\). Among functional monomers on the market, 10-methacryloyloxydecyl dihydrogen phosphate (10-MDP) stands out because it promotes chemical adhesion with hydroxyapatite and forms stable ionic bonds with dentin in an aqueous environment\(^\text{[32]}\). Despite its use in self-etching DBS, it has been a relevant ingredient in universal DBS.

Therefore, lasers can be used with universal DBS considering that they could facilitate the penetration and involvement of these agents and aid in their incorporation in modified substrates. Ayar and Erdemir\(^\text{[10]}\) investigated the adjunctive use of a universal bonding system with Er, Cr: YSGG on the enamel surface with interesting performance.

The purpose of this study was to analyze the effect of an MDP-based universal adhesive irradiated with Er, Cr: YSGG on dentin bonding over the course of 6 months.

**MATERIALS AND METHODS**

**Experimental design**

This *in vitro* study involved three major factors: the DBS in four levels (Adper Scotchbond\textsuperscript{TM} Universal [SU]–10-MDP based; Clearfil\textsuperscript{TM} SE Bond [CSE]–10-MDP based; Adper\textsuperscript{TM} Scotchbond Multipurpose [MP]–methacrylate-based, and Adper\textsuperscript{TM} Single Bond 2 [SB]–methacrylate-based); treatment in two levels (No irradiation–Control [C] and Er, Cr: YSGG laser irradiation [E]), and time (initial [I] and 6 months [6m]). The quantitative response variable was the bond strength determined using micro tensile test. The failure mode was also analyzed using portable digital microscopy at a 40x magnification.

**Specimen preparation**

A total of 80 sound human third molars were extracted for surgical reasons after consent and approval by the local Ethical Committee (protocol number 49812415.1.0000.5417), and were randomized using Excel software (Microsoft Office\textsuperscript{®}, Redmond, WA, USA) according to the dimensions of exposed dentin area into eight groups (n=10). Teeth were stored in 0.1% tymol solution. After the teeth were completely cleaned, the crowns were transversely sectioned on the occlusal third to expose the dentin using a low-speed diamond saw (Isomet\textsuperscript{TM} Low Speed Saw\textsuperscript{®}, Buebher, Lake Bluff, IL, USA) with a water-cooled diamond disc (Extec, Enfield, CA, USA). The remaining enamel was removed using #320 grit silicon carbide paper (Carbimet Paper Discs, Buehler) on a polishing machine (Arotec, Cotia, SP, Brazil). The samples were polished for 30 s with a #600 grit silicon carbide paper to simulate smear layer formation and stored in deionized water at 37 °C until the moment of DBS application.

**DBS application and laser treatment**

Two etch-and-rinse adhesives, one self-etch adhesive, and one universal adhesive were tested. All DBS were applied according to the manufacturers’ instructions (Table 1). For the etch-and-rinse adhesives, 37% phosphoric acid (Dentsply, Catanduva, SP, Brazil) was applied for 15 s, rinsed for the same time, and the dentin was then dried with absorbent paper (wet technique). For the MP and CSE, a single coat of primer was applied, and an air stream was gently applied for 5 s at a standardized distance of 5 cm. Subsequently, one coat of adhesive was applied, and the excess was removed. For SB, two coats of primer/bond were applied, and an air stream was gently applied for 5 s at a distance of 5 cm. For the universal adhesive (SU), a single coat of adhesive was actively applied for 20 s, the excess was removed, and an air stream was gently applied for 5 s at a standardized distance of 5 cm. All adhesives were light-cured for 10 s using an LED Blue Star 2 (1,000 mWcm\textsuperscript{2}, Microdont, São Paulo, SP, Brazil) device.

The Er, Cr: YSGG laser (Water Lase iPlus, Biolase, Irvine, CA, USA) was activated after primer application at a distance of 3 mm from the surface substrate (non-contact mode), 90\(^\circ\) inclination, and standardized by XY table after automatic scanning of the area (BioPDI, São Carlos, SP, Brazil). The table facilitated an automatic zigzag scan, where the displacement of the X axis was determined by the extent of the test area and that of the Y axis was based on the thickness of the optical fiber tip (Fig. 1). The time was standardized according to the largest specimen so that the entire dentin area of all specimens received irradiation with the same energy density. The standardized time between the beginning of the irradiation until the moment of polymerization was 60 s. For control groups, it was waited 60 s before light-curing to standardize the time between all specimens. The detailed parameters used for laser irradiation are listed in Table 2.

Lastly, the specimens were restored using Filtek Z250 (3M ESPE, St Paul, MN, USA) in three consecutives 1.5 mm increments. The specimens were stored in deionized water at 37°C for 7 days. Subsequently, the specimens were perpendicularly sectioned to the occlusal dentin with a low-speed diamond saw (Isomet, Buehler) to obtain resin-dentin sticks with 0.64 mm\textsuperscript{2} average cross-sectional dimensions. Twenty-five sticks were obtained from each tooth, approximately. The sticks were divided for the initial and 6-month groups. During the 6 months of storage, the sticks were kept in deionized water at 37 °C that was replaced every 15 days.

**Micro tensile bond strength test**

The specimens were tested using an Instron 3342 universal testing machine (Illinois Tool Works,
<table>
<thead>
<tr>
<th>Dentin Bonding</th>
<th>Manufacturer</th>
<th>Classification/ Lot number</th>
<th>Composition</th>
<th>Application technique</th>
</tr>
</thead>
<tbody>
<tr>
<td>Adper™ Single Bond 2</td>
<td>3M ESPE</td>
<td>Two-step Etch-and-rinse system/ N695410</td>
<td>HEMA*; Bis-GMA**, ethanol; silane treated silica filler; glycerol 1.3 dimethacrylates; diuretanedimethacrylate and copolymer of polyacrylic and polyitaconic acids.</td>
<td>1. Etch dentin surface with phosphoric acid 37% for 15 s&lt;br&gt;2. Rinse for 15 s&lt;br&gt;3. Gently dry with absorbent paper&lt;br&gt;4. Apply two coats of the primer/bond with a microbrush and remove excess&lt;br&gt;5. Gently air-dry for 5 s&lt;br&gt;6. Laser irradiation (for the irradiated groups)</td>
</tr>
<tr>
<td>Clearfil SE™ Bond</td>
<td>Kuraray, Okayama, Japan.</td>
<td>Two-step self-etch system/ 9N0169</td>
<td>Primer: MDP***; HEMA**; hydrophilic aliphatic dimethacrylate; dl-Camphorquinone; N,N-Diethanol-p-toluidine; Water. Bond: MDP***; HEMA*; Bis-GMA***; hydrophobic aliphatic dimethacrylate; dl-Camphorquinone; N,N-Diethanol-p-toluidine; colloidal silica.</td>
<td>1. Apply primer with a microbrush for 20 s&lt;br&gt;2. Gently air-dry for 5 s&lt;br&gt;3. Laser irradiation (for the irradiated groups)&lt;br&gt;4. Apply bond with a microbrush and remove excess</td>
</tr>
<tr>
<td>Adper™ Scotchbond™ Universal</td>
<td>3M ESPE, Seefeld, Germany.</td>
<td>Universal system (used as one-step self-etch system with wet-technique)/ 643238</td>
<td>MDP***; HEMA*; Bis-GMA**; silica treated silane; ethanol; decamethylene dimethacrylate; water; 1,10-decanediol dimethacrylate; copolymer of polyacrylic and polyitaconic acids; Camphorquinone; N,N-dimethylbenzoic acid; methacrylate 2-dimethylmonoethyl; methyl ethyl ketone.</td>
<td>1. Apply bond actively for 20 s&lt;br&gt;2. Gently air-dry for 5 s&lt;br&gt;3. Laser irradiation (for the irradiated groups)</td>
</tr>
</tbody>
</table>

*HEMA: 2-hydroxyethyl methacrylate
Bis-GMA: Bisphenol A diglycidyl methacrylate
MDP: 10-methacryloyloxydecyl dihydrogen phosphate
Fig. 1 Standardization design of the trajectory and irradiation time based on the largest area specimen (A). Therefore, all the specimens could receive the same energy density throughout the dentin area (B).

Table 2 Parameters used for irradiation

<table>
<thead>
<tr>
<th>Parameter</th>
<th>Value</th>
</tr>
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<tbody>
<tr>
<td>Energy per pulse (output)</td>
<td>25 mJ</td>
</tr>
<tr>
<td>Frequency</td>
<td>10 Hz</td>
</tr>
<tr>
<td>Power</td>
<td>0.25 W</td>
</tr>
<tr>
<td>Energy density</td>
<td>20.83 J/cm²</td>
</tr>
<tr>
<td>Thickness tip</td>
<td>800 µm</td>
</tr>
</tbody>
</table>

Norwood, IL, USA). The cross-sectional area of each stick was measured with a digital caliper (Digimatic Caliper Absolute, Mitutoyo, Kawasaki, Japan) and the values were logged into the onboard BlueHill software (BlueHill® Materials Testing Software, Norwood, IL, USA). Subsequently, the sticks were individually attached with a cyanoacrylate-based adhesive (Loctite Super Bonder Gel Control, Henkel, São Paulo, SP, Brazil) to the machine’s dispositive (JIG 1 Micro tensile, Odeme, Luzerna, SC, Brazil). The adhesive interface was positioned perpendicular to the tensile forces generated by the testing machine. Tension was applied at crosshead speed of 0.5 mm/min, with maximum load of 500 N that measured the force required to test the stick in Newtons (N).

Failure mode analysis

Both segments of the fractured specimens were evaluated to define the type of failure with a portable digital microscope (Dino Lite Microscope Plus, AnMo Electronics, New Taipei City, Taiwan) at 40× magnification. The specimens were classified by failure modes: adhesive (A); cohesive in dentin (CD); cohesive in resin (CR); and mixed (M), and the percentage of each failure type was obtained.

The methodology used is in accordance with the Guidelines proposed by the Academy of Dental Materials for micro tensile bond strength test. In each specimen, the fracture area in the both segments of the specimen were estimated. To be categorized as adhesive failure, the adhesive interface must present a separation between the segments for at least 50% of the area. For classification as cohesive failures in dentin or resin, the interface was not easily notable and one of these cohesive failures must be present at least about 50% of the area. The fracture was considered mixed if presented more than one type of fracture.

Representative images of each type of fracture were obtained using scanning electronic microscopy (SEM). The fractured sticks were fixed in stubs, coated with gold, and examined via SEM (JSM T220A, JEOL) at a magnification of ×75.

Statistical analysis

The obtained data was calculated and analyzed statistically with Statistica software (Statsoft®, Tulsa, OK, USA). The assumptions of normal distribution and equality of variances were checked for all the variables using Kolmogorov-Smirnov and Levene tests, respectively. As the assumptions were satisfied, the data were subjected to three-way ANOVA ($p<0.05$) followed by Tukey’s test ($p<0.05$) for individual comparisons.

RESULTS

The mean bond strengths and standard deviations are listed in Table 3. The data revealed statistical significance for all tested factors ($p<0.0001$). In addition, the interactions between the factors were significant ($p<0.0001$) and the interaction of all factors showed a $p$ value of 0.0194.

For the DBS, initial similar bond strength was observed regardless of pretreatment. When comparison was performed for each DBS with respect to time, all systems treated with no irradiation (control) exhibited stabilized performance. Except for SU, all groups treated with laser irradiation exhibited reduced bond strength after 6 months.

The initial bond strength was similar for all DBSs regardless of pretreatment. However, the 6-month analyses showed that laser irradiation significantly impaired the bond strength of all DBSs except for SU.

For all groups, the adhesive and mixed failure modes were the most common. A description of the distribution is presented in Fig. 2 and representative images of the fractures are shown in Fig. 3 (a to d).
### Table 3  Mean and standard deviations values (MPa) of bond strength

<table>
<thead>
<tr>
<th>DBS</th>
<th>Immediate</th>
<th>6 months</th>
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</thead>
<tbody>
<tr>
<td></td>
<td>Control</td>
<td>Laser</td>
</tr>
<tr>
<td>MP</td>
<td>34.23 (2.22) Aa*</td>
<td>30.43 (2.67) Aa*</td>
</tr>
<tr>
<td>SB</td>
<td>39.43 (2.74) Aa*</td>
<td>42.59 (6.03) Ab*</td>
</tr>
<tr>
<td>CSE</td>
<td>36.13 (4.27) Aa*</td>
<td>38.93 (2.27) Aab*</td>
</tr>
<tr>
<td>SU</td>
<td>42.45 (4.87) Aa*</td>
<td>46.62 (4.49) Ab*</td>
</tr>
</tbody>
</table>

n=10, p<0.05

Different uppercase letters indicate statistical differences for the same condition (DBS and pretreatment) in different time evaluation.

Different lowercase letters indicate statistical differences among the DBSs in the same pretreatment and time.

Different symbols indicate statistical differences for the same DBS and time regarding pretreatment.

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**DISCUSSION**

Universal adhesives were introduced to the market with the goal of simplifying the adhesive technique and reducing technical sensitivity\(^{34}\). However, studies have shown that this new generation of adhesive exhibits different physicochemical properties, resulting in morphological changes in the substrate and improving the longevity of the adhesive interface\(^ {7,35}\). To improve its performance, this study investigated the role of the adjunctive use of a multimode universal DBS with Er, Cr: YSGG laser irradiation, as previous studies indicated that this strategy can improve long-term use\(^ {15-17}\).

In terms of DBS, the highest bond strength was obtained for SU, showing stable performance over the investigational period. In a systematic review and meta-analysis by Rosa et al.\(^ {30}\), the studies that used the same brands, similar modes of self-etching, and storage time exhibited bond strength values similar to those obtained in this study.

Taking control pretreatment into account, all DBS showed stability in the bond strength to dentin when no laser irradiation was applied. However, when the laser was used, a drastic reduction in all conditions after 6 months was observed, except for the SU group.

The best performance for SU with time and pretreatment can be partially attributed to its composition, the phosphate acidified functional
monomer 10-MDP that forms a stable chemical bond with hydroxyapatite. This monomer creates stable MDP-calcium salts and a nanolayer interface between hydroxyapatite and MDP that exhibits multifunctional properties, such as durability of nanolayering. This protects the hybrid layer due to its hydrophobic nature, prevents collagen degradation, and promotes hydroxyapatite strength to acid dissolution.

Despite the promising results presented by the SU, the same did not observed with the CSE, a DBS that also has MDP in its composition. One of the differences between SU and CSE is the presence of the polyalkenoic acid copolymer in SU, commercially known as Vitrebond copolymer, which facilitates the incorporation of the DBS to the hydroxyapatite in the collagen scaffold. This difference deserves further investigation. In addition to these two chemical bonding components, SU is a simplified adhesive and contains hydroxyethyl methacrylate (HEMA) in the same bottle, which may prevent interfacial self-assembly. These factors were investigated by Tsijumoto et al. who studied the interfacial characteristics and bond durability of universal adhesives to various substrates. The authors observed that universal adhesives modify the interfacial characteristics of the substrates and create a consistent surface on the dentin.

Ceballo et al. investigated a similar laser (Er: YAG – 2,940 nm), in order to condition the dentin and improve the penetration of the adhesive. The results determining that its interaction could result in a poorly attached hybrid layer to the sound dentin below the irradiated area. Other study has shown that the Er, Cr: YSGG laser can affect the hydroxyl ion of hydroxyapatite, causing morphological changes in the structure such as the creation of a rough surface without a smear layer, open dentinal tubules, fissures, and cracks of peritubular dentin.

Although erbium lasers have similar wavelengths, studies with the Er, Cr: YSGG laser have shown better interaction with the hydroxyl ion (OH) presented in hydroxyapatite. The maximum absorption peak of the hydroxyl ion is approximately 2,800 nm, while for water is slightly wider (approximately 2,940 nm). Thus, studies suggest that this peak slightly closer to the Er, Cr: YSGG laser wavelength may reflect a better interaction of this laser compared to Er: YAG.

The use of Er, Cr: YSGG has been shown to be an additional method to improve the durability of universal DBS when applied to dentin, but some factors remain unclear since most previous studies did not involve this specific category of adhesive. More recent investigations of the role of the entrapment and chemical reaction of MDP with hydroxyapatite have proposed interesting mechanisms, in which laser interacts with both water and hydroxyapatite, two important components in the establishment of dental bonding.

In the present study, Er, Cr: YSGG laser irradiation was performed on dentin previously impregnated with an adhesive system. The authors speculated that laser treatment, which presents the similar wavelength of water and hydroxyapatite absorption, could eliminate the solvents and water present in the substrate with increasing surface temperature. Some additional studies should be performed to prove this hypothesis. However, based on the results presented herein, no immediate influence on the bond strength values between the control and laser groups was observed. This conflicts with previous studies that used other lasers with different wavelengths and the same technique but obtained an improved adhesive interface. The controversial result of this study may be because all these studies evaluated the dentin bonding interface immediately after treatment. Despite the promising results regarding SU, questions remain and require further investigation.

Another main finding of this study is regarding the MDP-based materials, which can improve the degree of conversion, indicating that it could minimize the progression of water induced degradation over time.

For the failure analysis, the methodology used is in accordance with the guidelines proposed by the Academy of Dental Materials. The most common failure in all groups were adhesive and mixed. For the groups with reduced bond strength, the adhesive failure increased. These results show that procedures that seek to increase the bond strength are interesting, one that reduces the risk of failure in the adhesive interface, increasing the retention of the resin clinically. The obtained results were according to the findings in the literature, validating this analysis.

CONCLUSIONS

This study was designed to determine the effect of a universal adhesive on sub-ablative Er, Cr: YSGG irradiated dentin. Based on the results presented herein, it can be concluded that:

The SU was capable of maintaining bond strength after 6 months for the substrate tested.

The sub-ablative laser technique applied on SU did not impair bonding as observed for all other categories of DBS.

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