Effect of Silicon on the Interaction between Recrystallization and Precipitation in Niobium Microalloyed Steels

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The effect of Si addition on the interaction between recrystallization and precipitation was investigated in terms of the no-recrystallization temperature ($T_{nr}$) on three microalloyed steels containing about 0.035 mass% Nb. The $T_{nr}$ was measured using torsion testing over the Si concentration range from 0.01 to 0.48 mass%. It was observed that the $T_{nr}$ increased with Si level, but appeared to saturate at long interpass times. In addition, high strains reduced the influence of Si on the $T_{nr}$. This behaviour is attributed to the acceleration of Nb(C, N) precipitation by the addition of Si.

KEY WORDS: recrystallization; strain-induced precipitation; no-recrystallization temperature ($T_{nr}$); silicon addition; niobium microalloyed steel; hot deformation; HSLA steel.

1. Introduction

The appropriate employment of microalloying elements in high strength low alloy (HSLA) steels, coupled with thermomechanical processing, can provide improvements in both strength and toughness.1–5) This is achieved by suitable manipulation of the recrystallization and precipitation phenomena that take place during deformation. Microalloying elements are present in HSLA steels as both solutes and precipitates. By suppressing austenite recrystallization, these elements act as ferrite grain refiners, thus increasing the yield strength and decreasing the impact transition temperature.

Carbonitride precipitation, especially strain-induced precipitation, depends largely on the temperature and deformation conditions, but can also be influenced by other alloying additions. Some non-precipitating elements, such as Mn and Si, can alter the activities of C and N5–10); that is to say, these elements can modify the effective solubility of Nb(C, N) and thereby influence the interaction between recrystallization and precipitation.

A clear understanding of this interaction involves the no-recrystallization temperature ($T_{nr}$), the temperature below which recrystallization cannot be completed within the interpass time range of a given multipass rolling operation. This temperature, which depends on the degree of supersaturation, is determined by the occurrence of precipitation; the latter in turn suppresses the progress of recrystallization.11) In addition to composition, the $T_{nr}$ is a function of strain, strain rate, interpass time and reheat temperature.12–16) An increase in the $T_{nr}$ allows finish rolling to be conducted at higher temperatures, which in turn enable lower loads to be developed or larger reductions to be applied without consuming additional energy.14) Furthermore, the critical parameter for shape and dimensional control during rolling is the rolling load, i.e. mean flow stress, which determines the roll gap. Therefore, predicting the $T_{nr}$ is important because it is associated with significant changes in mean flow stress.11)

The aim of this study, in terms of the $T_{nr}$, was to examine the effect of Si addition on the interaction between recrystallization and precipitation in a series of Nb microalloyed steels. This work seeks to develop further understanding of the mechanisms responsible for this effect.

2. Experimental Procedure

Three Nb-bearing steels containing Si concentrations of 0.01, 0.11 and 0.48 mass% were studied in the present work; their detailed chemical compositions are displayed in Table 1. Multipass torsion tests, which are widely used to simulate industrial hot rolling, were employed to determine the $T_{nr}$. The as-received plates were machined into torsion specimens, with their longitudinal axes aligned along the

<table>
<thead>
<tr>
<th>Table 1.</th>
<th>Chemical compositions of the steels (in mass%).</th>
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<tr>
<td></td>
<td>C</td>
</tr>
<tr>
<td>0.01Si</td>
<td>0.08</td>
</tr>
<tr>
<td>0.11Si</td>
<td>0.11</td>
</tr>
<tr>
<td>0.48Si</td>
<td>0.09</td>
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rolling direction. Each specimen had a gauge length of 22.4 mm and a diameter of 6.3 mm.

The experiments were conducted using a servo-hydraulic, computer controlled MTS torsion machine. For the purpose of preventing oxidation, the specimen and parts of the loading bars were contained in a quartz tube sealed with O-rings, through which a constant flow of high purity argon was passed.

The test schedule is illustrated schematically in Fig. 1 and the test parameters are displayed in Table 2. In order to avoid the occurrence of dynamic recrystallization, low pass strains were adopted. It appears from previous research 17,18 that strain rate does not have a significant influence on either the recrystallization kinetics or the $T_{nr}$. Thus, in the present work, all tests were conducted at a fixed strain rate of 2 s$^{-1}$. For a particular test, the pass strain and interpass time were held constant. For short interpass times ($t_{ip} < 10$ s), precipitation is unlikely to take place in the early stages of multipass deformation; only solute drag retards recrystallization. Therefore, the interpass times ($t_{ip}$) adopted in this investigation ranged from 20 to 200 s. Interpass times of 20 s are representative of reversing mills, such as plate or roughing mills; thus the present results are particularly applicable to this type of mill. The temperature decrease during each interpass interval was 30°C; therefore the average cooling rates ($\Psi$) in the different tests were given by

$$\Psi = 30°C/t_{ip}$$

In order to reveal the prior austenite grain boundaries, sections were cut perpendicular to the radius and at a depth of 0.1$r$ of the torsion samples. These were then polished and etched in saturated aqueous picric acid at 60–80°C containing several drops of hydrochloric acid and a wetting agent (detergent). The grain size was evaluated by the linear intercept method, as described in section E112 of the ASTM standards.

### Table 2. Test parameters for the multipass torsion tests.

<table>
<thead>
<tr>
<th>Strain/pass</th>
<th>$\varepsilon = 0.2, 0.35$</th>
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<tr>
<td>Strain rate</td>
<td>$\dot{\varepsilon} = 2 s^{-1}$</td>
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<tr>
<td>Interpass times</td>
<td>$T_p = 20, 30, 60, 100, 200 s$</td>
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</table>

### 3. Results

#### 3.1. Determination of the $T_{nr}$

The equivalent stresses and strains were calculated from the torque and twist data using the following equations, as employed by earlier workers:

$$\sigma_{eq} = \frac{3.3\sqrt{3}I}{2\pi^3}, \quad \varepsilon_{eq} = \frac{r\Theta}{3L}$$

Here, $I$ is the measured torque, $\Theta$ is the measured angle of twist, $r$ is the specimen radius, $L$ is the gauge length. Figure 2 shows some of the typical stress–strain curves obtained. The mean flow stress (MFS) for each pass was calculated in terms of the equivalent stress and strain by numerical integration; that is:

$$MFS = \bar{\sigma}_{eq} = \frac{1}{\varepsilon_{eq} - \varepsilon_{eq}} \int_{\varepsilon_{eq}}^{\varepsilon_{eq} \delta\varepsilon_{eq}}$$

The MFS's are plotted as a function of inverse absolute temperature in Fig. 3. Based on the slope change that can be seen, the continuous increase in flow stress can be divid-
ed into two regions. In region I, recrystallization takes place fairly rapidly, so that there is no accumulation of work hardening. The increase in MFS can only be attributed to the temperature drop and is thus small. In region II, the rate of change of MFS increases perceptibly. This is because only partial or no recrystallization occurs, so that the work hardening is accumulated from pass to pass. In this temperature range, the increase in MFS is due, not only to the decrease in temperature, but also to the retention of work hardening. According to Boratto et al., the temperature corresponding to the slope change at the intersection of regions I and II can be defined as the no-recrystallization temperature ($T_{nr}$).

When deformation is carried out at lower temperatures, the austenite-to-ferrite transformation occurs (region III), leading to a drop in mean flow stress because ferrite is softer than austenite. In region IV, where this transformation is more or less complete, the mean flow stress increases again as a result of both strain hardening of the ferrite–pearlite structure and the decrease in temperature. The temperatures that correspond to the start and finish of this transformation can be defined as the $\text{Ar}_3$ and $\text{Ar}_1$.

### 3.2. Austenite Microstructure

The deformation microstructures were examined with the aim of determining when the retardation of recrystallization occurred due to precipitation. Water quenching was performed on the 0.01Si and 0.48Si steels, according to the test schedule displayed in Fig. 4. The initial austenite structures for these materials are shown in Fig. 5. The $T_{nr}$'s measured for the 0.01Si and 0.48Si steels corresponding to these conditions were 938°C and 962°C, respectively. The quenching temperature used was 940°C. For the 0.01Si steel, this final temperature is above its $T_{nr}$, meaning that recrystallization was completed before Nb(C, N) precipitation could be initiated. Therefore, the initially large austenite grains, with an average size of 365 µm, were refined to around 50 µm (see Fig. 6(a)) by repeated recrystallization. However, for the 0.48Si steel, the final temperature was

![Fig. 4. Test schedule for the 0.01Si and 0.48Si steels, showing quenching point.](image)

![Fig. 5. Reheated austenite microstructures; soaking temperature=1230°C, holding time=15 min.](image)

![Fig. 6. Austenite grain structures in (a) 0.01Si steel and (b) 0.48Si steel; $t_q=30$s, $\varepsilon=0.35$ and $\dot{\varepsilon}=2$s$^{-1}$, specimens quenched at 940°C.](image)
below its $T_{nr}$; that is to say, Nb(C, N) precipitation started before quenching occurred. Thus the equiaxed grains were gradually replaced by pancaked grains, as illustrated in Fig. 6(b); this was due to suppression of the progress of recrystallization by precipitation.

### 3.3. Effect of Interpass Time

A set of MFS vs. $10^4/T$ curves for the 0.11Si steel is illustrated in Fig. 7. For these tests, the pass strain (0.35/pass) and strain rate ($2 \text{ s}^{-1}$) were held constant and the interpass time was altered between tests (from 20 to 200 s). When deformation proceeds below the $T_{nr}$ the kinetics of recrystallization are retarded, whilst precipitation is accelerated. The MFS values for interpass times of 30 s are, therefore, higher than those for 20 s. However, with further increases in the interpass time, precipitate coarsening is initiated at temperatures close to the nose of the precipitation-temperature-time (PTT) curve, leading to the reduced influence of precipitation. This explains the relative positions of the MFS curves for interpass times of 30 s and >30 s in Fig. 7; here, the 60 s, 100 s and 200 s MFS values are generally lower than those for interpass times of 30 s at temperatures below the $T_{nr}$.

The effect of interpass time on the $T_{nr}$ for the 0.11Si steel is displayed in Fig. 8. For times ranging from 20 to 200 s, the diagram can be divided into two regions: precipitation and precipitate coarsening. In the precipitation region, the $T_{nr}$ increases with interpass time until, after a certain time (for instance, around 30 s for the 0.11Si steel), the $T_{nr}$ starts to decrease (precipitate coarsening). The dependence of the $T_{nr}$ on interpass time in the 0.01Si and 0.48Si steels is similar to that of the 0.11Si steel, as can be seen in Fig. 8. Further interpretation of these behaviors is given in Sec. 4.

### 3.4. Effect of Pass Strain

Two MFS vs. $10^4/T$ plots are displayed in Fig. 9; these demonstrate the effect of pass strain on the $T_{nr}$ for the 0.48Si steel. These particular specimens were tested at a constant interpass time of 20 s. It can be seen from the diagram that the MFS decreases with increasing pass strain. In addition, the point at which the slope change of the MFS vs. inverse temperature curve occurs moves to lower temperatures when the pass strain is increased. The influence of pass strain on the $T_{nr}$ for the 0.48Si steel is displayed in Fig. 10. The results show that the $T_{nr}$ drops with increasing pass strain in all cases.

### 3.5. Effect of Silicon Content

The mean flow stress vs. $10^4/T$ curves for the steels with the three different Si contents are presented in Fig. 11 at a similar pass strain (0.2/pass) and interpass time (20 s).
It can be seen that the mean flow stress increases with Si concentration from 0.11 to 0.48 mass%. The rate of increase, however, varies greatly with interpass time and pass strain, as shown in Fig. 12. When relatively short interpass times were used, involving the early stages of precipitation, the increase in $T_{nr}$ with Si content was evident. These times correspond to those conventionally employed in reversing mills. However, this effect seemed to saturate at longer interpass times. Under these conditions, there is no additional increase in $T_{nr}$ with Si content. Such long interpass times are not encountered in typical rolling schedules. In addition, it should be noted that the extent of the increase in $T_{nr}$ with Si level decreases when the pass strain is raised from 0.2 to 0.35, as displayed in Fig. 12.

4. Discussion

4.1. Interaction between Recrystallization and Precipitation

Interpass time has an important effect on the interaction between recrystallization and precipitation as it determines whether the precipitation of microalloy carbonitrides is possible and which type of recrystallization (static, metadynamic, etc.) can take place. Previous researchers have shown that the presence of dislocations promotes the nucleation, growth and coarsening of precipitates. Dutta et al. proposed that nucleation takes place at dislocation nodes in the three-dimensional dislocation network introduced by deformation, as illustrated in Fig. 13. Due to the acceleration of solute diffusion between particles by pipe diffusion, coarsening can readily take place. According to Dutta et al., coarsening starts at an early stage of precipitation and thereby leads to a substantial decrease in the precipitate number density. In addition, precipitate growth and coarsening take place simultaneously during precipitation.

The results of the present work agree very well with this model. As can be seen from Fig. 8, in the precipitation region, recrystallization is retarded to a significant degree by the occurrence of strain-induced precipitation and the $T_{nr}$ increase.
increases with interpass time. Raising the interpass time increases the volume fraction of precipitates, thus enhancing the retardation of recrystallization by particle pinning until the interpass time reaches a critical value. At longer interpass times, particle coarsening leads to a decrease in the grain boundary pinning force; recrystallization can therefore continue to proceed until new precipitates provide additional pinning at lower temperatures. Thus, the $T_{nr}$ decreases with increased interpass time in the precipitate coarsening region. The interaction between recrystallization and precipitation is also strongly dependent on strain. It is well known that the time required for 50% recrystallization to occur, $t_{0.5r}$, decreases with an increase in strain, through a relationship that can be expressed as:

$$t_{0.5r} \propto e^{-m} \quad \text{(4)}$$

The value of the coefficient $m$, according to Fernandez et al.,\cite{25} varies between 2 and 4 for austenite grain size ranges of 20–1000 µm in steels with chemical compositions similar to those of the present work. The start time for strain-induced recrystallization is also reduced by increasing the strain. A relationship between the precipitation start time ($t_{0.05p}$) and strain has been determined for Nb microalloyed steels\cite{26} to be:

$$t_{0.05p} \propto e^{-1} \quad \text{(5)}$$

The rates of change of $t_{0.5r}$ and $t_{0.05p}$ with strain can be expressed by the first partial derivatives of Eqs. (3) and (4):

$$\frac{\partial t_{0.5r}}{\partial e} \propto -me^{-m+1} \quad m = 2–4 \quad \text{(6)}$$

$$\frac{\partial t_{0.05p}}{\partial e} \propto -e^{-2} \quad \text{(7)}$$

Since dislocations can be nucleation sites for both recrystallization and precipitation, the increase in dislocation density resulting from increased strain accelerates the rates of both processes. Nevertheless, as displayed in Fig. 10, the $T_{nr}$ decreases as the strain is increased. This is because the rates of acceleration for the two phenomena are different. By comparing Eqs. (5) and (6), it can be seen that the precipitation kinetics are less sensitive to strain than the recrystallization kinetics, although this difference is gradually reduced as the strain approaches unity. Furthermore, an increase in dislocation density also leads to more rapid particle coarsening. When these coarsened particles lose their effectiveness in pinning the grain boundaries, recrystallization is able to proceed at lower temperatures.

### 4.2. Effect of Silicon

Figure 12 clearly demonstrates that the influence of Si on raising the $T_{nr}$ is not constant for different interpass times and pass strains. There are two mechanisms that can be responsible for this effect.

#### a) Solute Drag Effect

Bai and co-workers\cite{27} demonstrated that, even in the absence of Nb(C, N) precipitation, a $T_{nr}$ can still be measured at short interpass times as a result of the inhibition of recrystallization by Nb solute drag. In their recent investigation regarding the effect of Si on austenite recrystallization, Serajzadeh and Taheri\cite{28} observed that increasing the Si concentration in plain carbon steels can increase the incubation time and decrease the growth rate of static recrystallization.

On the other hand, Mavropoulos and Jonas\cite{29} observed that Mn in solution has an almost negligible direct solute drag effect on the migration of grain boundaries. Similarly, in the work of Akben et al.,\cite{30} it was shown that a Nb HSLA steel containing 1.90 mass% Mn had a slightly lower peak strain in dynamic recrystallization than a steel of similar composition, but containing only 1.25 mass% Mn. A reasonable explanation could be that, although Mn and Si do not make direct contributions to solute drag, they may exert indirect effects by influencing the diffusivity of Nb in austenite.\cite{31} A somewhat similar explanation has been proposed in the work of Boratto et al.\cite{32-34} They postulated that increasing the Si content has a negative effect on the $T_{nr}$ because Si decreases the solubility of Nb(C, N) in austenite, thereby leading to a decrease in the Nb solute drag.

#### b) Pinning Effect due to Strain-induced Precipitation

In addition to the key role of site density in determining the precipitation kinetics of heterogeneous nucleation, the effects of alloying elements such as Si and Mn on the solubility of Nb are also important. Mn decreases the rate of Nb(C, N) precipitation,\cite{35} in contrast to the effect of Si. This is because Si addition increases the apparent concentrations of C and N, and therefore the apparent solubility product of Nb(C, N).\cite{36} It is well known that the occurrence of precipitation is affected by the supersaturation level of Nb(C, N), which increases as the temperature is decreased during deformation. High levels of supersaturation increase the driving force for precipitation and thereby enable precipitation to start at higher temperatures. This suppresses recrystallization at higher temperatures, which in turn increases the $T_{nr}$.

The increase in $T_{nr}$ with Si level can either be caused by an increase in solute Si content in the matrix or by the enhanced precipitation of Nb(C, N). The solute Si drag effect is, however, relatively small compared with the inhibition produced by the strain-induced precipitation of Nb(C, N).\cite{37-39} Therefore, Si is not considered as severe a recrystallization inhibitor as Nb, even though it has an effect on the kinetics of recrystallization when present in the absence of solute Nb.\cite{40} Once the strain-induced precipitation of Nb(C, N) takes place, the effect of Si in solid solution can be neglected. In other words, when Si coexists with Nb in steel, its influence on the progress of recrystallization is not directly attributable to the presence of Si in solution, but via its effect on the kinetics of Nb(C, N) precipitation.

If the effect of Si in raising the $T_{nr}$ were due to the presence of Si in solution in the matrix, then the influence of Si on the $T_{nr}$ would not vary with interpass time. When Si coexists with niobium in steel, the tendency for Si to form carbidies or nitrides is small, due to the strong affinity of Nb for C and N. Therefore, whatever the interpass time, the solute Si content in the matrix should remain constant. However, from the appearance of the $T_{nr}$ curve displayed in Fig. 12, it can be seen that the decrease in the $T_{nr}$ with Si content varies for different interpass conditions. As mentioned in the previous sections, increased interpass...
4.3. Effect of Silicon Addition in the Presence of Other Carbide Forming Elements

Synergistic effects have also been observed when Si and V are both present in a steel in the absence of other carbide forming elements. \( \text{Si} \) decreases the solubility of Nb carbide in both austenite and ferrite by raising the C activity. However, when Si coexists with other carbide-forming elements, some additions have an influence opposite to that of Si, as they decrease the activity of C. Thus, when other carbide-forming elements (such as Cr and Mo) coexist with Si, the overall combination can lead to little or no effect on the solubility of Nb carbide or Nb carbonitride. It has been reported that in V steels containing large amounts of other carbide-forming additions, no effect of Si on V precipitation was observed. Based on the view that Si has similar effects on the solubilities of both Nb and V carbide, it is expected that Si would have no effect on the \( T_{np} \) in Nb steels containing large amounts of other carbide-forming elements.

5. Conclusions

(1) Over a certain interpass time range \((20 \text{s} \leq t_p \leq 200 \text{s})\), the \( T_{np} \) increases with Si content. That is to say, in steels containing both Si and Nb additions, precipitation is capable of retarding recrystallization at higher temperatures than in steels containing only Nb.

(2) In opposition to the conclusions drawn from previous research on microalloyed steels, where the Nb remained in solution, the positive effect of Si addition on the \( T_{np} \) is due to the indirect effect of Si on the kinetics of Nb(C,N) precipitation.

(3) The increase in \( T_{np} \) associated with Si addition is not constant under different deformation conditions. When the interpass times are short \((20 \text{s} \leq t_p \leq 30 \text{s})\), as in reversing mills, the \( T_{np} \) increases with interpass time and Si content. The effect of Si on the \( T_{np} \) appears to saturate at very long interpass times; the latter are beyond those employed industrially.

(4) Increasing the pass strain appears to accelerate the recrystallization kinetics more than the precipitation kinetics. This may be because an increase in dislocation density leads to more rapid coarsening of the precipitates. As a result, the increase in \( T_{np} \) associated with Si addition decreases at high pass strains.

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