Simultaneous Determination of Azimsulfuron, Flazasulfuron and Halosulfuron-methyl in Grains, Seeds, Vegetables and Fruits by HPLC

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A method was developed for the simultaneous determination of 3 sulfonylurea herbicides, azimsulfuron, flazasulfuron and halosulfuron-methyl in agricultural products. The herbicides were extracted with acetone, re-extracted with ethyl acetate, and then transferred to 2% dipotassium hydrogenphosphate solution. The herbicides were extracted again into ethyl acetate and cleaned up using Sep-Pak® Plus Alumina N and Bond Elut® SAX cartridge columns. The 3 sulfonylurea herbicides were determined by HPLC. The fortified peaks were confirmed by LC/MS with electrospray ionization (ESI), and the peaks of azimsulfuron, flazasulfuron and halosulfuron-methyl were determined. The recoveries of the 3 sulfonylurea herbicides from brown rice, corn, cotton seed, ginkgonut, chestnut, almond, walnut, cucumber, pumpkin, orange, grapefruit, mandarin, lemon and grape ranged from 77.0 to 112.3% following fortification at 0.05–0.5 µg/g. The detection limits were 0.01 µg/g for azimsulfuron and halosulfuron-methyl, and 0.02 µg/ml for flazasulfuron (S/N > 3).

Key words —— sulfonylurea herbicide, HPLC, Sep-Pak® Plus Alumina N, Bond Elut® SAX

INTRODUCTION

Sulfonylurea herbicides are a relatively new class of compounds whose herbicidal activity was discovered in the mid-1970s. They show high activity at low application levels and low mammalian toxicity; at least 16 sulfonylurea herbicides are currently on the market around the world. The Ministry of Health, Labour and Welfare (MHLW) in Japan has already specified the tolerance levels for residual sulfonylurea herbicides such as azimsulfuron, bensulfuron-methyl, chlorimuron-ethyl, chlorosulfuron, flazasulfuron, imazasulfuron, metsulfuron-methyl and tribenuron-methyl. The MHLW intends to set the maximum residue limits (MRL) and the official analytical method for halosulfuron-methyl in agricultural products. The official analytical methods for pesticides under the Food Sanitation Law are mostly individual determinations. We have attempted the simultaneous determination of 9 sulfonylurea herbicides including halosulfuron-methyl; however, some of them showed low recoveries, which may be caused by their low solubility in aqueous solution or insufficient recoveries from the cleanup columns. Few studies have reported on the analysis of multiple sulfonylurea herbicides in agricultural products by HPLC-photodiode array detection and capillary electrophoresis.

The MHLW has already set the MRL in agricultural products for 229 pesticides under the Food Sanitation Law. Among them, we have reported the HPLC determination of the official analytical method for emamectin and clethodim in agricultural products.

In the present study, the simultaneous determination of azimsulfuron and flazasulfuron, which have been in effect since November 22, 1999, and halosulfuron-methyl in agricultural products was conducted. Recently, the MHLW has set the MRL and the official analytical method for azimsulfuron, flazasulfuron and halosulfuron-methyl in agricultural products.

MATERIALS AND METHODS

Samples —— Brown rice, corn, ginkgonut, chestnut, almond, walnut, cucumber, pumpkin, orange, grapefruit, mandarin, lemon and grape were purchased from markets in Osaka, Japan. The cotton seeds were provided by MHLW.

Reagents ——

Acetone, acetonitrile, n-hexane, methanol and ethyl acetate: Pesticide residue analytical grade (Wako Pure Chemical Industries, Ltd., Osaka, Japan).
Hydrochloric acid, dipotassium hydrogen-phosphate, diatomaceous earth Celite 545, sodium chloride, trichloroacetic acid and sodium sulfate: Special grade (Wako Pure Chemical Industries).

Sodium sulfate: Activated at 120°C for 12 hr.

Standard materials: Azimsulfuron, flazasulfuron and halosulfuron-methyl were obtained from Wako Pure Chemical Industries. The chemical structures of these compounds are shown in Fig. 1.

Pesticide standard solution: Standard solutions of azimsulfuron, flazasulfuron and halosulfuron-methyl (1000 µg/ml) were prepared by dissolving each pesticide in acetonitrile. For the recovery experiments, the standard solution was diluted with acetonitrile (10 µg/ml).

HPLC and liquid chromatography/mass spectrometry (LC/MS): Mobile phase A, 0.01% trichloroacetic acid in distilled water; mobile phase B, acetonitrile.


HPLC and LC/MS Analysis ——

Apparatus: The HPLC and LC/MS used in our previous study.8)

Operating parameters of HPLC: The mobile phase flow rate was adjusted to 1.0 ml/min during the analysis. The system was equilibrated at 20% mobile phase B in mobile phase A; a 15 min linear gradient to 100% mobile phase B was begun and held for 5 min. When the gradient was completed, the mobile phase was returned to 80% A, 20% B and held for 7 min to re-equilibrate the column. The other conditions were as follows: temperature for column separation, 40°C, and ultraviolet detection wavelength, 245 nm.

Operating parameters of LC/MS: The stainless steel column (2.0 mm i.d. × 100 mm) was packed with Tosoh TSKgel Super-ODS. The mobile phase flow rate was adjusted to 0.2 ml/min during the analysis. The system was equilibrated at 20% mobile phase B in mobile phase A; a 12-min linear gradient to 100% mobile phase B was begun and held for 3 min. The other chromatographic conditions were as described for the operating parameters of the HPLC.

MS conditions: Analytical mode, electrospray ionization (ESI negative); drying gas (N2) flow, 4.5 l/min; probe voltage, −4.5 kV. The selected ion for monitoring was m/z 423 for azimsulfuron, 406 for flazasulfuron and 433 for halosulfuron-methyl.

Extraction —— Ten g of shredded sample (brown rice, corn, cotton seed, ginkgonut, chestnut, almond, walnut) was placed in a stainless steel cup to which 20 ml of water was added and allowed to stand for 2 hr, and then 100 ml of acetone was added. Twenty g of chopped cucumber, pumpkin, orange, grapefruit, mandarin, lemon or grape was placed in a stainless steel cup to which 100 ml of acetone was added. The mixture was homogenized for 3 min and then filtered through filter paper with 7 g of Celite 545 (10 mm thickness) into a 300-ml round-bottom flask. The extract was rinsed, filtered with 50 ml acetone, and evaporated to dryness with a rotary evaporator. The extract was transferred to a 200-ml beaker, to which was added 100 ml of 10% sodium chloride solution and then adjusted to pH 3–4 with 1 mol/l of
hydrochloric acid. The extract was transferred to a 300-ml separatory funnel, 50 ml ethyl acetate added, and vigorously shaken for 5 min. Another 50 ml of ethyl acetate was added, and the solution was shaken again for 5 min. The organic layers were collected in a 300-ml separatory funnel, 100 ml of n-hexane and 50 ml of 2% dipotassium hydrogenphosphate solution added, and vigorously shaken for 5 min. Another 50 ml of 2% dipotassium hydrogenphosphate solution was added, and the solution was again shaken for 5 min. The aqueous layers were collected in a 200-ml beaker and then adjusted to pH 3–4 with 6 mol/l of hydrochloric acid. The extract was transferred to a 300-ml separatory funnel, 50 ml of ethyl acetate added, and vigorously shaken for 5 min. Another 50 ml of ethyl acetate was added, and the solution was again shaken for 5 min. The organic layers were collected in a 300-ml Erlenmeyer flask, dehydrated with 20 g of anhydrous Na$_2$SO$_4$, and allowed to stand for 30 min. They were then filtered through filter paper to separate the anhydrous Na$_2$SO$_4$.

The flask was then rinsed with an additional 20 ml of ethyl acetate and evaporated to dryness under vacuum at 40°C.

**Cleanup** ———

*Procedure A:* A Sep-Pak® Plus Alumina N cartridge column was conditioned with 10 ml of acetonitrile before use. The residue was dissolved in 5 ml of acetonitrile and charged onto the column. The column was rinsed with 10 ml of acetonitrile, followed by elution with 15 ml of 20% water in acetonitrile, and the eluate was evaporated to dryness under vacuum at 40°C. The residue was dissolved in 30 ml of ethyl acetate and transferred to a 100-ml Erlenmeyer flask, dehydrated with 20 g of anhydrous Na$_2$SO$_4$, and allowed to stand for 30 min. The solution was then filtered through filter paper to separate the anhydrous Na$_2$SO$_4$. The flask was then rinsed with an additional 20 ml of ethyl acetate, and filtered through filter paper. The solution was evaporated to dryness under vacuum at 40°C.

*Procedure B:* A Bond Elut® SAX cartridge column was conditioned with 10 ml of acetonitrile in acetone before use. The residue was dissolved in 5 ml of 25% n-hexane in acetone and charged onto the column. The column was rinsed with 10 ml of 25% n-hexane in acetone, followed by elution with 15 ml of 10% methanol in acetone. The eluate was evaporated to dryness under vacuum at 40°C and dissolved in 1 ml of acetonitrile.

**Quantification** ——— The sample solution was automatically injected into the HPLC system for residue analysis. The concentration of azimsulfuron, flazasulfuron and halosulfuron-methyl was calculated based on a peak area calibration curve. The injection was performed three times for each sample to test the reproducibility.

**Recovery Test** ——— Chopped samples were fortified with 0.05–0.5 µg/g azimsulfuron, flazasulfuron and halosulfuron-methyl and analyzed by the proposed method. The recovery data represent 5 replications.

**RESULTS AND DISCUSSION**

HPLC Chromatogram, Linearity and Limit of Detection

An HPLC chromatogram of the standards is shown in Fig. 2. The retention times of azimsulfuron, flazasulfuron and halosulfuron-methyl are 10.6, 11.5 and 13.0 min, respectively. In the present study, trichloroacetic acid was used as the ion-pair reagent, whereas 1-heptanesulfonic acid could not be used because azimsulfuron, flazasulfuron and halosulfuron-methyl were not eluted from the column.

The linear dynamic range of the detector response at 245 nm for azimsulfuron, flazasulfuron and halosulfuron-methyl was examined and appeared to be from 0.05 to 4 µg injected on-column. The detection limits were 0.01 µg/ml for azimsulfuron and
halosulfuron-methyl and 0.02 µg/ml for flazasulfuron (S/N > 3).

**Extraction**

Azimsulfuron, flazasulfuron and halosulfuron-methyl in agricultural products were extracted with acetone.

The official analytical method for azimsulfuron and flazasulfuron⁹ used 2% dipotassium hydrogenphosphate solution for the extraction from ethyl acetate into the aqueous layer. However, the recovery of halosulfuron-methyl was approximately 50%, which may be due to its low solubility in aqueous solution. The result is shown in Table 1. The addition of 100 ml of n-hexane to ethyl acetate layer before shaking with 2% potassium hydrogenphosphate solution allowed for good recovery of the halosulfuron-methyl. This result indicated that the simultaneous determination of the 3 sulfonylurea herbicides prepared by the official analytical method for azimsulfuron and flazasulfuron⁹ was employed with slight modification.

**Cleanup**

The official analytical method for azimsulfuron and flazasulfuron⁹ used an alumina open column and Bond Elut® SAX for sample purification. The alumina open column and Sep-Pak® Plus Alumina N cartridge column were compared for recoveries and cleaning up the extract. Almost the same recoveries were obtained, and the use of the alumina open column is more time-consuming for sample preparation. However, the cleanup was not satisfactory with this column; therefore, a second cleanup with another column was necessary. We evaluated the use of ion exchange cartridge columns such as Bond Elut® SAX, PRS, NH2, SCX, and PSA cartridge columns for sample purification. The Bond Elut® SAX cartridge column gave the best recovery among them. This result indicated that the same column gave satisfactory recoveries with the official analytical method for azimsulfuron and flazasulfuron⁹ and our analytical method.

Typical chromatograms of an almond-fortified 0.2 µg/g (A) sample of the three pesticides and an almond blank (B) are shown in Fig. 3. The HPLC chromatogram of the sample solution of cotton seed shows an interfering peak close to the retention time of flazasulfuron after cleanup with 2 cartridge col-

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**Table 1. Transfer of Sulfonylurea Herbicides from Ethyl Acetate Layer to Potassium Hydrogenphosphate Layer by the Addition of n-Hexane**

<table>
<thead>
<tr>
<th>Added volume of n-hexane(ml)</th>
<th>Azimsulfuron</th>
<th>Flazasulfuron</th>
<th>Halosulfuron-methyl</th>
</tr>
</thead>
<tbody>
<tr>
<td>0 ml</td>
<td>98.8 ± 3.4</td>
<td>97.3 ± 2.8</td>
<td>50.0 ± 3.1</td>
</tr>
<tr>
<td>50 ml</td>
<td>97.1 ± 3.8</td>
<td>96.6 ± 3.3</td>
<td>72.7 ± 3.8</td>
</tr>
<tr>
<td>100 ml</td>
<td>96.5 ± 3.3</td>
<td>95.3 ± 3.4</td>
<td>97.7 ± 3.5</td>
</tr>
</tbody>
</table>

a) Average ± standard deviation of 5 determinations. A 0.5 µg portion of each sulfonylurea herbicide was added to 100 ml of ethyl acetate and some n-hexane and 50 ml of 2% potassium hydrogenphosphate solution and shaken vigorously for 5 min. Another 50 ml of 2% potassium hydrogenphosphate solution was added, and the solution was again shaken for 5 min. The aqueous layers were collected and then adjusted to pH 3–4 with 6 mol/l of hydrochloric acid. The aqueous layer was added to 50 ml of ethyl acetate and shaken vigorously for 5 min. Another 50 ml of ethyl acetate was added, and the solution was shaken again for 5 min. The organic layers were collected, evaporated to dryness under vacuum at 40°C, and dissolved in 1 ml of acetonitrile.

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**Fig. 3. Typical HPLC Chromatograms of Agricultural Product Samples**

Peaks: 1 = Azimsulfuron; 2 = flazasulfuron; 3 = halosulfuron-methyl. Chromatograms are almond-fortified to 0.2 µg/g (A) and the almond blank (B).
UV detection is commonly used for pesticide residue analysis by HPLC. However, an agricultural product sample shows many interference peaks in the UV detection. The detected peaks in the samples were then confirmed by LC/MS (ESI). The selected ion modes for monitoring were $m/z$ 423 for azimsulfuron, $m/z$ 406 for flazasulfuron and $m/z$ 433 for halosulfuron-methyl. Figure 4 shows an almond-fortified 0.2 µg/g (A) sample and an almond blank (B) by LC/MS (SIM). The peak of azimsulfuron, flazasulfuron and halosulfuron-methyl in almond blank was not detected. The result indicated that...

Fig. 4. Selected Ion Monitoring (SIM) Chromatograms of Agricultural Product Samples by LC/MS (ESI) Peaks: 1 = Azimsulfuron; 2 = flazasulfuron; 3 = halosulfuron-methyl. SIM chromatograms are almond-fortified to 0.2 µg/g (A) and the almond blank (B).
when measuring pesticides in agricultural products including many interfering peaks on UV detection, using the SIM of LC/MS significantly improved the qualitative analyses.

**Recovery Test**

The recoveries of azimsulfuron, flazasulfuron and halosulfuron-methyl in 14 agricultural products fortified at 0.05–0.5 µg/g are shown in Table 2. In Japan, the common acceptable range of recovery for pesticide residue is 70–120%. The recoveries of azimsulfuron, flazasulfuron, and halosulfuron-methyl were 77.0–92.3%, 83.3–112.3% and 86.3–95.0%, respectively. The CV of the recovery was within 10%.

### REFERENCES


### Table 2. Recoveries of Azimsulfuron, Flazasulfuron and Halosulfuron-methyl Added to Agricultural Products

<table>
<thead>
<tr>
<th>Sample</th>
<th>Fortified level (µg/g)</th>
<th>Azimsulfuron (mean ± S.D.)</th>
<th>Flazasulfuron (mean ± S.D.)</th>
<th>Halosulfuron-methyl (mean ± S.D.)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Brown rice</td>
<td>0.05</td>
<td>84.0 ± 5.3</td>
<td>87.3 ± 5.0</td>
<td>90.7 ± 3.1</td>
</tr>
<tr>
<td>Corn</td>
<td>0.05</td>
<td>83.7 ± 7.1</td>
<td>90.7 ± 5.5</td>
<td>95.0 ± 1.7</td>
</tr>
<tr>
<td>Cotton seed</td>
<td>0.05</td>
<td>88.3 ± 3.5</td>
<td>112.3 ± 2.5</td>
<td>94.7 ± 3.1</td>
</tr>
<tr>
<td>Ginkgonut</td>
<td>0.05</td>
<td>88.3 ± 3.8</td>
<td>88.7 ± 4.0</td>
<td>93.0 ± 1.7</td>
</tr>
<tr>
<td>Chestnut</td>
<td>0.05</td>
<td>92.3 ± 4.0</td>
<td>85.3 ± 4.0</td>
<td>92.3 ± 1.5</td>
</tr>
<tr>
<td>Walnut</td>
<td>0.05</td>
<td>86.0 ± 5.3</td>
<td>87.7 ± 1.5</td>
<td>87.0 ± 5.3</td>
</tr>
<tr>
<td>Almond</td>
<td>0.2</td>
<td>85.0 ± 5.2</td>
<td>86.3 ± 1.2</td>
<td>89.7 ± 1.5</td>
</tr>
<tr>
<td>Cucumber</td>
<td>0.5</td>
<td>80.9 ± 1.3</td>
<td>83.3 ± 4.4</td>
<td>88.4 ± 4.3</td>
</tr>
<tr>
<td>Pumpkin</td>
<td>0.5</td>
<td>77.0 ± 3.5</td>
<td>83.3 ± 4.4</td>
<td>90.6 ± 4.1</td>
</tr>
<tr>
<td>Orange</td>
<td>0.1</td>
<td>87.7 ± 5.1</td>
<td>89.0 ± 4.0</td>
<td>92.3 ± 2.5</td>
</tr>
<tr>
<td>Grapefruit</td>
<td>0.1</td>
<td>83.0 ± 4.4</td>
<td>87.7 ± 4.5</td>
<td>88.7 ± 5.1</td>
</tr>
<tr>
<td>Mandarin</td>
<td>0.1</td>
<td>81.0 ± 2.0</td>
<td>83.4 ± 4.5</td>
<td>91.0 ± 5.3</td>
</tr>
<tr>
<td>Lemon</td>
<td>0.1</td>
<td>80.0 ± 2.0</td>
<td>83.3 ± 4.2</td>
<td>88.7 ± 2.1</td>
</tr>
<tr>
<td>Grape</td>
<td>0.1</td>
<td>89.0 ± 1.0</td>
<td>90.3 ± 3.8</td>
<td>86.3 ± 2.5</td>
</tr>
</tbody>
</table>

**a)** Average ± standard deviation of 5 determinations.