Cuprobismutite group minerals (cuprobismutite, hodrušhite, kupčíkite and paděraite), other Bi–sulfosalts and Bi–tellurides from the Obari mine, Yamagata Prefecture, Japan

Yuya IZUMINO*, Kazuo NAKASHIMA** and Mariko NAGASHIMA***

*Graduate School of Science and Engineering, Yamagata University, Kojirakawa 1-1-4, Yamagata 990-8560, Japan
**Department of Earth and Environmental Sciences, Yamagata University, Kojirakawa 1-4-12, Yamagata 990-8560, Japan
***Graduate School of Science and Engineering, Yamaguchi University, Yoshida 1677-1, Yamaguchi 753-8512, Japan

Cuprobismutite group minerals have been discovered for the first time in Japan from the Obari mine, Yamagata Prefecture. Furthermore, many kinds of Bi–sulfosalts and Bi–tellurides have been found from this mine. Cuprobismutite occurs as irregular particles which are intergrown with makovickyite. The empirical formula of cuprobismutite is (Cu7.71Fe0.41)∑24.12(Bi12.17Ag1.57Pb0.07Sb0.07)∑13.88(S23.95Se0.06)∑24.01 (based on total atoms = 46). Hodrušhite, kupčíkite and paděraite occur as euhedral to subhedral blade-shaped crystals with parallel intergrowth. The empirical formulae of these minerals are (Cu7.64Fe0.36)∑22.07(Bi11.18Cu0.35Ag0.26Pb0.07Sb0.07)∑11.93 (S21.99Se0.09)∑22.07 (based on total atoms = 42), (Cu3.56Fe0.50Zn0.04)∑24.10(Bi4.84Sb0.03)∑13.87(S10.00Se0.03)∑10.03 (based on total atoms = 19) and Cu7.00[(Cu0.50Ag0.17)∑0.67Pb1.27Bi11.01Sb0.05]∑12.03 (S21.91Se0.09)∑22.00 (based on total atoms = 42) for hodrušhite, kupčíkite and paděraite, respectively. The content of trace elements (Ag, Fe and Pb) in each cuprobismutite group mineral is consistent with the characteristics shown by previous works.

**Keywords:** Cuprobismutite, Hodrušhite, Kupčíkite, Paděraite, Obari mine

INTRODUCTION

The Obari mine, located in Tsuruoka City, Yamagata Prefecture, Japan is famous in the country as a high-grade Bi-deposit. This mine is also known for the occurrence of Bi–sulfosalts such as aitkite (Takaoka, 1969) and wittichenite (Watanabe, 1937, 1938a, 1938b, 1938c, 1951, Sugaki et al., 1974), however for a long time, the other Bi–sulfosalts have not been studied in detail. Recently, makovickyite and cupromakovickyite were discovered during a re-investigation of Bi–bearing ores from the mine (Izumino et al., 2013). Moreover, the crystal structure of makovickyite from this mine was studied in detail by Nagashima et al. (2013). During these studies, we have found many kinds of Bi–sulfosalts and Bi–tellurides including cuprobismutite group minerals (cuprobismutite, hodrušhite, kupčíkite and paděraite) for the first time in Japan. This paper mainly deals with the mode of occurrence and chemistry of the cuprobismutite group minerals as well as other Bi–sulfosalts and Bi–tellurides.

Cuprobismutite group minerals, belonging to sulfosalts of Cu–Bi–S systems, have been classified into cuprobismutite homologous minerals (cuprobismutite, hodrušhite and kupčíkite) and paděraite as a species related to the series (Makovicky, 1989). The former is represented by the general formula, CuM2N−1−2Bi4S2N−8, where M is Bi and Ag (Makovicky, 1989). Cuprobismutite with an ideal formula of Cu3AgBi13S24, is the most major mineral of cuprobismutite group minerals and was first described from the Missouri deposit, USA (Hillebrand, 1884). Hodrušhite with an ideal formula of Cu3Bi12S22, often occurs in the world and was first reported from the Banská Hodruša, Romania (Koděra et al., 1970). Kupčíkite with an ideal formula of Cu3Fe0.33Bi5S10, is a rare mineral and was found in Bătaia Bihor, Romania for the first time (Mumme...
and Žák, 1985). The crystal structures of all these minerals were refined by Topa et al. (2003a) and Topa and Makovicky (2006). The group minerals mainly consist of Cu, Bi and S, with minor components (Fe, Ag and Pb), maintaining the charge balance. In general, cuprobismutite contains significant amounts of Ag, hodrušhite includes small amounts of Fe and Ag, kupčíkite has a high content of Fe, and paděraite contains substantial amounts of Pb. The crystal structure analyses of the minerals indicated interdependent substitutions such as (1) Cu ↔ Bi, (2) Ag ↔ Bi, (3) Pb ↔ Bi and (4) Fe ↔ Cu (Topa et al., 2003a; Topa and Makovicky, 2006).

**GEOLOGICAL SETTING**

Late Cretaceous–Paleogene granitoids are distributed in the studied area and most of the granitoids belong to the ilmenite series (Ishihara et al., 1983). The geology around the Obari mine is composed of Cretaceous Tagawa acidic rocks, Cretaceous to Paleogene Wasada granodiorite and Neogene sedimentary rocks of the Oizumi formation (Fig. 1). Tagawa acidic rocks consist of dacitic pyroclastic rocks and granodiorite porphyry, and are partly metamorphosed by the intrusion of the Wasada granodiorite (Yauchi et al., 1973). The Wasada granodiorite belongs to exceptionally the magnetite series, and has many Au–Ag–Cu–Pb–Zn–Bi–Mo deposits in and around the granitic body. The Obari mine is the largest ore deposit in this region. This mineralization is related to the Wasada granodiorite (Agency of Natural Resources and Energy, 1982). The Oizumi Formation consists of volcanic pyroclastic rocks and volcanic rocks, and is widely distributed in covering unconformably the Tagawa acidic rocks and Wasada granodiorite (Agency of Natural Resources and Energy, 1982).

The Obari mine is an Au–Cu–Bi dissemination vein type deposit which is altered along the fissure in the Wasada granodiorite and was mined during 1936–1966. The dissemination vein is characterized by potassic alteration (Takaoka, 1969), and K–feldspar, muscovite and quartz were formed in the altered host rock. Main ore minerals are chalcopyrite, tetrahedrite and Bi–sulfosalts, and are disseminated in the gangue minerals. The grade of ore, Au: 200 g/t, Ag: 150–500 g/t, Cu: 10–15%, Bi: 3–8% (Shimazu et al., 1961), indicates that the Bi concentration is much higher than in other Bi deposits in Japan.

**ORE MINERALOGY**

Ore samples used in this study were obtained by a former employee of the Obari mine. Ore minerals composed of the specimens are Bi–sulfosalts, Bi–tellurides, chalcopyrite, pyrite, sphalerite, galena, tetrahedrite, tennantite, bornite, arsenopyrite, molybdenite, electrum and hessite. Gangue minerals are quartz, muscovite, K–feldspar, albite, rutile, anatase, brookite, apatite and zircon. Among the ore minerals except for Bi minerals, chalcopyrite is the most common mineral in this mine. The mineral occurs as massive aggregates and often coexists with aikinite (Fig. 2a) and wittichenite (Fig. 2b). Bornite is closely associated with chalcopyrite, and includes lamellae of chalcopyrite and wittichenite (Fig. 2c). Tetrahedrite–tennantite replaces crystals of chalcopyrite (Fig. 2d) and bornite. Sphalerite sometimes occurs in chalcopyrite and contains chalcopyrite dots in the grains. Galena is closely associated with cosalite and aikinite. Electrum occurs as small inclusions in aikinite and chalcopyrite (Fig. 2e). Arsenopyrite and pyrite show euhedral crystals and are partly replaced by chalcopyrite. Molybdenite disseminates as small tabular crystals in chalcopyrite. Hessite rarely occurs as small inclusions within wittichenite.

Bi–sulfosalts and Bi–tellurides discovered in this study are cuprobismutite group minerals (cuprobismutite, hodrušhite, kupčíkite and paděraite), makovickyite, cupromakovickyite, wittichenite, emplèctite, miharaite, aikinite–bismuthinite series (aikinite, friedrichite, hammarite, lindströmite, krupkaite, pekoite and bismuthinite), galenobismutite, cannizzarite, cosalite, lillianite, nuffiel-
dite, tellurobismuthite, tsumoite, tetradymite, joséite-B, and rucklidgeite. In addition to these minerals, native Bi and secondary minerals such as bismutite occur in the ores. Makovickyite and cupromakovickyite have been described in detail by Izumino et al. (2013).

**METHOD**

Chemical analyses were carried out using a JEOL JXA8600S wavelength dispersive electron microprobe at the Faculty of Science, Yamagata University. Analytical conditions for quantitative analysis are accelerating voltage, 20 kV; sample current, 20 nA; beam diameter, 3 µm, and correction, ZAF. The standards used were CuFeS₂ (CuKα, FeKα and SKa), GaAs (AsLα), Ag (AgLα), Sb₂S₃ (SbLα), ZnS (ZnKα), PbS (PbMα), Cd (CdLα), Bi (BiLα), Se (SeLα) and Te (TeLα). Standard deviations (error in wt%) of elements for the cuprobismutite group minerals are: Ag 0.05, Cu 0.11, Fe 0.05, Pb 0.05, Bi 0.69, Sb 0.04, Se 0.04 and S 0.08. The results of EPMA analyses of Bi-sulfosalts and Bi-tellurides are plotted on the ternary diagrams (Fig. 3) and are presented in Table 1. The cuprobismutite group minerals are too small and too closely associated with each other to be identified using X-ray diffraction.

To distinguish between cosalite and neyite, X-ray diffraction data on single crystals of cosalite were collected using a Bruker SMART APEX II CCD diffractometer installed at the University of Bern, Switzerland. A crystal of cosalite was mounted on a glass fiber and intensity data was measured at room temperature using graphite-monochromatized MoKα radiation (λ = 0.71069 Å). Preliminary lattice parameters and an orientation matrix were obtained from twelve sets of frames and refined during the integration process of the intensity data. Diffraction data were collected with ω scans at different ϕ settings (ϕ–ω scan) (Bruker, 1999). Data were processed using SAINT (Bruker, 1999).
Figure 3. Ternary diagrams of Bi-sulfosalts and Bi-tellurides: (a) Cuprobismutite group minerals, (b) Cu-Bi-S minerals, (c) Akinie-bismuthinite series, (d) Pb-Bi-S minerals and (e) Bi-tellurides.
Table 1. Representative chemical compositions of Bi-sulfosalts and Bi-tellurides from the Obari mine

<table>
<thead>
<tr>
<th>Mineral name</th>
<th>Ag</th>
<th>Cu</th>
<th>Fe</th>
<th>Zn</th>
<th>Cd</th>
<th>Pb</th>
<th>Bi</th>
<th>Sb</th>
<th>As</th>
<th>Se</th>
<th>Te</th>
<th>S</th>
<th>Total</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Cuprobismutite group minerals</strong></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Cuprobismutite (NA = 15)</td>
<td>4.22</td>
<td>12.18</td>
<td>0.58</td>
<td>0.01</td>
<td>0.02</td>
<td>0.36</td>
<td>63.23</td>
<td>0.20</td>
<td>0.01</td>
<td>0.11</td>
<td>0.02</td>
<td>19.09</td>
<td>100.03</td>
</tr>
<tr>
<td><strong>Hodrušite (NA = 30)</strong></td>
<td>0.78</td>
<td>14.10</td>
<td>0.56</td>
<td>0.02</td>
<td>0.00</td>
<td>0.38</td>
<td>64.88</td>
<td>0.24</td>
<td>0.01</td>
<td>0.19</td>
<td>0.01</td>
<td>19.56</td>
<td>100.72</td>
</tr>
<tr>
<td><strong>Kupčíkite (NA = 30)</strong></td>
<td>0.03</td>
<td>14.24</td>
<td>1.77</td>
<td>0.17</td>
<td>0.00</td>
<td>0.00</td>
<td>63.62</td>
<td>0.25</td>
<td>0.01</td>
<td>0.15</td>
<td>0.00</td>
<td>20.21</td>
<td>100.46</td>
</tr>
<tr>
<td><strong>Paděraine (NA = 30)</strong></td>
<td>0.48</td>
<td>12.67</td>
<td>0.03</td>
<td>0.00</td>
<td>0.00</td>
<td>7.02</td>
<td>61.15</td>
<td>0.16</td>
<td>0.00</td>
<td>0.19</td>
<td>0.06</td>
<td>18.66</td>
<td>100.42</td>
</tr>
<tr>
<td><strong>Cu-Bi-S minerals</strong></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Wittichenite (NA = 23)</td>
<td>3.47</td>
<td>37.14</td>
<td>0.03</td>
<td>0.04</td>
<td>0.00</td>
<td>0.00</td>
<td>39.73</td>
<td>0.06</td>
<td>0.00</td>
<td>0.02</td>
<td>0.04</td>
<td>19.58</td>
<td>100.12</td>
</tr>
<tr>
<td>Emplectite (NA = 25)</td>
<td>0.05</td>
<td>19.87</td>
<td>0.02</td>
<td>0.01</td>
<td>0.00</td>
<td>0.00</td>
<td>60.15</td>
<td>0.29</td>
<td>0.02</td>
<td>0.10</td>
<td>0.00</td>
<td>19.54</td>
<td>100.05</td>
</tr>
<tr>
<td>Aikinite-bismuthinite series</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Aikinite (NA = 17)</td>
<td>0.01</td>
<td>11.34</td>
<td>0.01</td>
<td>0.01</td>
<td>0.02</td>
<td>35.14</td>
<td>36.39</td>
<td>0.09</td>
<td>0.00</td>
<td>0.03</td>
<td>0.01</td>
<td>16.94</td>
<td>100.00</td>
</tr>
<tr>
<td>Friedrichite (NA = 6)</td>
<td>0.00</td>
<td>9.53</td>
<td>0.01</td>
<td>0.00</td>
<td>0.00</td>
<td>28.90</td>
<td>43.03</td>
<td>0.15</td>
<td>0.00</td>
<td>0.05</td>
<td>0.01</td>
<td>17.08</td>
<td>98.79</td>
</tr>
<tr>
<td>Hammarte (NA = 11)</td>
<td>0.00</td>
<td>7.47</td>
<td>0.01</td>
<td>0.00</td>
<td>0.00</td>
<td>23.47</td>
<td>49.25</td>
<td>0.88</td>
<td>0.00</td>
<td>0.03</td>
<td>0.01</td>
<td>17.64</td>
<td>98.77</td>
</tr>
<tr>
<td>Lindströmite (NA = 9)</td>
<td>0.00</td>
<td>6.96</td>
<td>0.01</td>
<td>0.01</td>
<td>0.02</td>
<td>21.75</td>
<td>51.92</td>
<td>0.58</td>
<td>0.00</td>
<td>0.02</td>
<td>0.02</td>
<td>17.71</td>
<td>99.00</td>
</tr>
<tr>
<td>Krupkaite (NA = 20)</td>
<td>0.00</td>
<td>6.45</td>
<td>0.02</td>
<td>0.01</td>
<td>0.01</td>
<td>19.21</td>
<td>56.68</td>
<td>0.23</td>
<td>0.00</td>
<td>0.03</td>
<td>0.03</td>
<td>17.81</td>
<td>100.62</td>
</tr>
<tr>
<td>Pekoite (NA = 7)</td>
<td>0.01</td>
<td>1.86</td>
<td>0.01</td>
<td>0.00</td>
<td>0.00</td>
<td>5.50</td>
<td>71.84</td>
<td>0.57</td>
<td>0.01</td>
<td>0.11</td>
<td>0.00</td>
<td>18.53</td>
<td>98.44</td>
</tr>
<tr>
<td><strong>Bismuthinite (NA = 23)</strong></td>
<td>0.01</td>
<td>1.25</td>
<td>0.01</td>
<td>0.00</td>
<td>0.00</td>
<td>3.47</td>
<td>75.60</td>
<td>0.33</td>
<td>0.01</td>
<td>0.20</td>
<td>0.01</td>
<td>18.83</td>
<td>99.70</td>
</tr>
<tr>
<td><strong>Pb-Bi-S minerals</strong></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Gallenobismuthite (NA = 8)</td>
<td>0.00</td>
<td>0.00</td>
<td>0.01</td>
<td>0.00</td>
<td>0.03</td>
<td>27.45</td>
<td>53.15</td>
<td>1.21</td>
<td>0.00</td>
<td>0.04</td>
<td>0.01</td>
<td>17.25</td>
<td>99.15</td>
</tr>
<tr>
<td>Cannizzarite (NA = 7)</td>
<td>1.17</td>
<td>0.45</td>
<td>0.15</td>
<td>0.00</td>
<td>0.08</td>
<td>34.15</td>
<td>46.29</td>
<td>0.76</td>
<td>0.00</td>
<td>0.26</td>
<td>0.30</td>
<td>16.52</td>
<td>100.12</td>
</tr>
<tr>
<td><strong>Cosalite (NA = 19)</strong></td>
<td>1.27</td>
<td>2.77</td>
<td>0.01</td>
<td>0.00</td>
<td>0.11</td>
<td>36.97</td>
<td>38.56</td>
<td>2.64</td>
<td>0.00</td>
<td>0.03</td>
<td>0.13</td>
<td>16.97</td>
<td>99.45</td>
</tr>
<tr>
<td><strong>Lillianite (NA = 19)</strong></td>
<td>1.05</td>
<td>0.14</td>
<td>0.01</td>
<td>0.00</td>
<td>0.05</td>
<td>46.53</td>
<td>35.66</td>
<td>0.56</td>
<td>0.00</td>
<td>0.10</td>
<td>0.15</td>
<td>16.03</td>
<td>100.27</td>
</tr>
<tr>
<td><strong>Other Bi-sulfosalts</strong></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Miharauite (NA = 11)</td>
<td>0.04</td>
<td>27.98</td>
<td>0.07</td>
<td>0.03</td>
<td>0.02</td>
<td>23.03</td>
<td>21.69</td>
<td>0.01</td>
<td>0.00</td>
<td>0.00</td>
<td>0.01</td>
<td>21.16</td>
<td>100.05</td>
</tr>
<tr>
<td>Nuffieldite (NA = 12)</td>
<td>0.04</td>
<td>7.19</td>
<td>0.03</td>
<td>0.00</td>
<td>0.03</td>
<td>36.49</td>
<td>36.79</td>
<td>2.93</td>
<td>0.00</td>
<td>0.13</td>
<td>0.04</td>
<td>17.11</td>
<td>100.79</td>
</tr>
<tr>
<td><strong>Bi-tellurides</strong></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Tetradymite (NA = 11)</td>
<td>0.06</td>
<td>0.13</td>
<td>0.00</td>
<td>0.00</td>
<td>0.00</td>
<td>0.25</td>
<td>58.48</td>
<td>0.27</td>
<td>0.00</td>
<td>0.00</td>
<td>35.06</td>
<td>4.82</td>
<td>99.08</td>
</tr>
<tr>
<td>Jodšite-B (NA = 9)</td>
<td>0.02</td>
<td>0.29</td>
<td>0.01</td>
<td>0.00</td>
<td>0.00</td>
<td>0.85</td>
<td>71.82</td>
<td>0.23</td>
<td>0.02</td>
<td>0.03</td>
<td>21.95</td>
<td>2.93</td>
<td>98.14</td>
</tr>
<tr>
<td>Tsumoite (NA = 11)</td>
<td>0.03</td>
<td>0.01</td>
<td>0.01</td>
<td>0.00</td>
<td>0.00</td>
<td>0.73</td>
<td>60.99</td>
<td>0.23</td>
<td>0.01</td>
<td>0.00</td>
<td>36.52</td>
<td>0.01</td>
<td>98.54</td>
</tr>
<tr>
<td>Rücklidgeite (NA = 2)</td>
<td>0.18</td>
<td>0.17</td>
<td>0.08</td>
<td>0.00</td>
<td>0.00</td>
<td>0.35</td>
<td>54.12</td>
<td>0.27</td>
<td>0.00</td>
<td>0.00</td>
<td>43.57</td>
<td>0.09</td>
<td>98.83</td>
</tr>
<tr>
<td>Tellurobismuthite (NA = 9)</td>
<td>0.33</td>
<td>0.01</td>
<td>0.01</td>
<td>0.00</td>
<td>0.00</td>
<td>0.08</td>
<td>51.05</td>
<td>0.28</td>
<td>0.00</td>
<td>0.00</td>
<td>47.46</td>
<td>0.03</td>
<td>99.27</td>
</tr>
</tbody>
</table>

NA, number of analyzed spots.
MODE OF OCCURRENCE AND CHEMICAL COMPOSITION

Cuprobismutite group minerals

Cuprobismutite is the rarest mineral of the cuprobismutite group minerals in the Obari mine, and occurs as irregular particles of 10 to 20 µm, and is intergrown with makovickyite and a Fe-bearing undetermined secondary Bismutinite (Fig. 4a). This mineral does not coexist with other cuprobismutite group minerals. The empirical formula of the mineral is \(\text{Cu}_{7.71}\text{Fe}_{0.41}\text{Bi}_{11.18}\text{Ag}_{0.35}\text{Pb}_{0.07}\text{Sb}_{0.07}\text{S}_{21.98}\text{Se}_{0.09}\) (based on total atoms = 42), corresponding to the ideal formula \(\text{Cu}_{7.64}\text{Fe}_{0.36}\text{Bi}_{11.93}\text{S}_{22.01}\text{Se}_{0.10}\). The chemical calculated N ranges between 1.77 and 2.06 (average: 1.92, ideally: 2). It contains significant amounts of Ag, small amounts of Fe and trace amounts of Pb, Sb and Se, and is rich in Cu and Fe and poor in Bi (Fig. 3a).

 Hodruštite, kupčíkite and paděráte occur as euhedral to subhedral blade-shaped crystals with parallel integrowth, and are accompanied by makovickyite, bismuthinite, kupčíkite and emplectite (Figs. 4b, 4c and 4d). This occurrence is characteristic for cuprobismutite group minerals and has been reported from other localities such as the Paulus mine, Romania (Cook and Ciobanu, 2003). Hodruštite is 10–250 µm in length and 5–30 µm in width. The empirical formula of the mineral is \(\text{Cu}_{7.54}\text{Fe}_{0.30}\text{Bi}_{11.18}\text{S}_{21.98}\text{Se}_{0.09}\) (based on total atoms = 42), corresponding to the ideal formula \(\text{Cu}_{7.64}\text{Bi}_{11.22}\). The chemically calculated N ranges between 1.20 and 1.39 (average: 1.27, ideally: 1.5). This mineral has small amounts of Ag and Fe and trace amounts of Pb, Sb and Se, and is rich in Bi and poor in Cu (Fig. 3a). Kupčíkite is 5–50 µm in length and 1–10 µm in width. The empirical formula of the mineral is \(\text{Cu}_{11.56}\text{Fe}_{0.50}\text{Ag}_{0.26}\text{Pb}_{0.07}\text{Sb}_{0.07}\text{S}_{10.03}\) (based on total atoms = 19), corresponding to the ideal formula \(\text{Cu}_{11.66}\text{Fe}_{0.56}\text{Bi}_{5.10}\). The chemically calculated N has values from 0.78 to 0.98 (average: 0.88, ideally: 1). It contains significant amounts of Fe and trace amounts of Zn, Sb and Se, and is rich in Cu, Fe and Zn and poor in Bi (Fig. 3a). Paděráte is 100–1000 µm in length and 1–20 µm in width. The empirical formula of the mineral is \(\text{Cu}_{27.00}\text{Fe}_{0.00}\text{Ag}_{18.00}\text{Bi}_{11.00}\text{Sb}_{0.00}\text{S}_{13.00}\text{S}_{9.09}\text{S}_{22.00}\) (based on total atoms = 42), corresponding to the ideal formula \(\text{Cu}_{27.20}\text{Fe}_{0.00}\text{Ag}_{18.00}\text{Bi}_{11.00}\text{S}_{13.00}\text{S}_{9.09}\text{S}_{22.00}\). This mineral has high contents of Pb and small contents of Ag, Sb, and Se, and is rich in Cu and poor in Bi (Fig. 3a). Each mineral can be distinguished using X-ray intensity mapping (Fig. 5) and the BSE images (Figs. 4c and 4d). In particular, hodruštite, kupčíkite and paděráte are distinguished by the difference in their minor components (Ag, Fe and Pb) of these minerals in X-ray intensity mapping (Fig. 5). Therefore, hodruštite contains small amounts of Fe and Ag, kupčíkite is the richest in Fe and paděráte has the highest Pb content.

Cu-Bi-S minerals

Wittichenite, the most common Bi-sulfosalt in the Obari mine, occurs as anhedral crystals of 0.1 to 3.0 mm in size, intergrown with aikinite, bornite, chalcopyrite and tetrahedrite (Fig. 2b). The eutectic intergrowths of bornite-chalcopyrite-wittichenite are observed in the studied samples (Fig. 2c), which have been studied by Watanabe (1938c), Sugaki (1949a), and Sugaki (1949b). The empirical formula is \(\text{Cu}_{4.16}\text{Fe}_{2.88}\text{Bi}_{0.94}\text{S}_{2.02}\) (based on total atoms = 7), corresponding to the ideal formula \(\text{Cu}_{4}\text{Bi}_{3}\text{S}_{5}\). This mineral contains significant amounts of Ag, up to 14.7 wt%, instead of Cu (Fig. 3b).

Emplecite occurs as irregular grains from 60 to 300 µm in size, coexisting with wittichenite and aikinite, with cuprobismutite group minerals and aikinite-bismuthinite series (Figs. 4b–4d) and with paděráte and aikinite. This mineral replaces cuprobismutite group minerals and the aikinite-bismuthinite series. The empirical formula is \(\text{Cu}_{1.03}\text{Bi}_{0.05}\text{Pb}_{0.97}\text{S}_{0.02}\) (based on total atoms = 4), corresponding to the ideal formula \(\text{CuBi}_{3}\). Aikinite, the second most common Bi-sulfosalt in the Obari mine, is widely seen, however, except for aikinite, the occurrences of the aikinite-bismuthinite series are rare. This mineral occurs as large anhedral crystals of 0.1 to 7 mm in size, accompanied by wittichenite, cosalite, chalcopyrite, bornite and tetrahedrite (Figs. 2a and 2b). The empirical formula is \(\text{Cu}_{1.02}\text{Pb}_{0.98}\text{Bi}_{1.00}\text{S}_{2.02}\) (based on total atoms = 6), corresponding to the ideal formula \(\text{CuPb}_{3}\).

Friedrichite, hammarite and lindströmite occur as irregular grains at the rim of cosalite and galenobismuthite (Fig. 4e). These minerals are not homogeneous and their compositions are dispersed around each ideal formula. The empirical formulae of these minerals are \(\text{Cu}_{10}\text{Pb}_{4.74}\) (based on total atoms = 35) for fridrichite, \(\text{Cu}_{11}\text{Pb}_{4.88}\) (based on total atoms = 17) for hammarite and \(\text{Cu}_{10}\text{Pb}_{2.88}\) (based on total atoms = 28) for lindströmite, respectively (Fig. 3c).

Krupkaite occurs as irregular particles from 10 to 100 µm in size at the boundary between bismuthinite and makovickyite (Fig. 4e). The empirical formula is \(\text{Cu}_{1.09}\).
Pb$_{0.99}$(Bi$_{2.91}$Sb$_{0.02}$)$_{2.93}$(S$_{5.96}$Se$_{0.02}$)$_{5.98}$ (based on total atoms = 11), corresponding to the ideal formula (CuPbBi$_3$S$_6$).

Pekoite rarely occurs as irregular particles of 10 to 30 µm in size, intergrown with hammarite in Type-C (Fig. 4e). The empirical formula is Cu$_{0.92}$Pb$_{0.84}$(Bi$_{10.96}$Sb$_{0.15}$)$_{11.11}$(S$_{18.26}$Se$_{0.04}$)$_{18.30}$ (based on total atoms = 31), corresponding to the ideal formula (CuPbBi$_{11}$S$_{18}$).

Bismuthinite occurs as irregular grains from 100 to 500 µm in size, coexisting with cuprobismutite group minerals from the Obari mine. 183
minerals, makovickyite and tetradyrmite (Fig. 4b). The empirical formula is Cu\(0.10\)Pb\(0.08\)(Bi\(1.83\)Sb\(0.01\))\(\sum\)1.84(S\(2.96\)Se\(0.01\))\(\sum\)2.97 (based on total atoms = 5), corresponding to the ideal formula (Bi\(_2\)S\(_3\)). This mineral has high contents of Cu and Pb (Fig. 3c).

Pb–Bi–S minerals

Galenobismutite occurs as irregular grains of up to 400 µm in size, as intergrowths with cosalite, hammarite and aikinite (Fig. 4e). The empirical formula is Pb\(0.99\)(Bi\(1.90\)Sb\(0.07\))\(\sum\)1.97S\(4.03\) (based on total atoms = 7), corresponding to the ideal formula (PbBi\(_2\)S\(_4\)).

“Cannizzarite” rarely occurs as euhedral to subhedral elongated crystals of 40 µm in size within bismuthinite (Fig. 4f). The empirical formula is (Ag\(_{0.47}\)Cu\(_{0.31}\)Fe\(_{0.12}\)Pb\(_{7.23}\))\(\sum\)8.13(Bi\(_{9.72}\)Sb\(_{0.27}\))\(\sum\)9.99(S\(_{22.62}\)Te\(_{0.10}\)Se\(_{0.15}\))\(\sum\)22.86 (based on total atoms = 41), corresponding to the ideal formula (Pb\(_8\)Bi\(_{10}\)S\(_{23}\)). Cannizzarite is not a single mineral, but a variable-fit homologous series (Topa et al., 2010), and some different cell-parameters have been reported from two localities, Vulcano, Italy (Topa et al., 2010) and Kudriavy volcano, Russia (Borisov et al., 2012).

Cosalite is the main Bi–sulfosalt of Pb–Bi–S systems in the Obari mine. The mode of occurrence of cosalite from the mine is divided into two types. One occurs as
large anhedral crystals of 0.1 to 3 mm in size, coexisting with aikinite, tetradyomite, galena and chalcopyrite in (Fig. 4g), and with galenobismutite and the aikinite-
bismuthnite series (Fig. 4e). The empirical formula is (Ag_{0.1}Cu_{0.4}Sb_{0.02})_{2.01}Bi_{1.97}Sb_{0.01}Te_{1.91}S_{1.06} (based on total atoms = 5). The obtained unit cell parameters are $a = 23.8405(11)$, $b = 4.0479(2)$, $c = 19.0436 \, \AA$, $V = 1837.78 \, \AA^3$, $Z = 8$. Although this mineral is rich in Ag and Cu and its chemical composition is similar to neyite, the mineral is identified as cosalite by the unit cell parameters. The crystal structure of Ag and Cu rich cosalite has been reported by Topa and Makovicky (2010). The other occurs as irregular grains of 10 to 100 \, \mu m in size in association with tetradyomite and lillianite (Fig. 4h). The empirical formula is (Ag_{0.01}Cu_{0.01}Bi_{3.89}Sb_{0.02})_{1.91} (S_{4.97}Te_{1.00}Se_{0.01})_{1.02} (based on total atoms = 11), corresponding to the ideal formula (PbBiS$_3$).

**Other Bi–sulfosalts**

Miharaite was originally reported from the Mihara mine (Sugaki et al., 1980), but the occurrence of this mineral is rare in Japan. This mineral from the Obari mine occurs as irregular particles from 1 to 30 \, \mu m in size as intergrowths with tetradymite and cosalite (Fig. 4i). The empirical formula is Cu$_4$FePbBiS$_6$. The other occurs as irregular particles from 10 to 20 \, \mu m in size, as intergrowths with tetradyomite and cosalite (Fig. 4j). The empirical formula is (Ag_{0.12}Cu_{0.03})_{2.01}Pb_{1.99}Bi_{1.95}S_{1.06} (based on total atoms = 9). The obtained unit cell parameters are $a = 23.8405(11)$, $b = 4.0479(2)$, $c = 19.0436 \, \AA$, $V = 1837.78 \, \AA^3$, $Z = 8$. Although this mineral is rich in Ag and Cu and its chemical composition is similar to neyite, the mineral is identified as cosalite by the unit cell parameters. The crystal structure of Ag and Cu rich cosalite has been reported by Topa and Makovicky (2010). The other occurs as irregular grains of 10 to 100 \, \mu m in size in association with tetradyomite and lillianite (Fig. 4h). The empirical formula is (Ag_{0.01}Cu_{0.01}Bi_{3.89}Sb_{0.02})_{1.91} (S_{4.97}Te_{1.00}Se_{0.01})_{1.02} (based on total atoms = 11), corresponding to the ideal formula (PbBiS$_3$).

**Bi–tellurides**

Tetradyomite is the most common Bi–telluride in the Obari mine and is seen in various Bi–mineral assemblages. The mode of occurrence of this mineral is divided into two types. One occurs as small irregular particles from 1 to 30 \, \mu m in size. The empirical formula is (Cu_{0.01}Pb_{0.01}Bi_{1.97}Sb_{0.02})_{2.01}Bi_{2.01}Te_{1.91}S_{1.06} (based on total atoms = 5). The oth-
er occurs as large tabular crystals of 0.1 to 2 mm in size, which has small inclusions of tellurobismutite, rucklidge-
ete and tsumoite, accompanied by lillianite and cosalite (Fig. 2f). The empirical formula is (Pb_{0.01}Bi_{1.97}Sb_{0.01})_{2.01}Te_{1.91}S_{1.07}. Both tetradymites correspond to the ideal for-

Joséite–B occurs as irregular grains of 40 to 100 \, \mu m in size, in association with tetradyomite and aikinite (Fig. 4k). The empirical formula is (Cu_{0.01}Pb_{0.01}Bi_{1.97}Sb_{0.01})_{2.01}Te_{1.91}S_{1.07} (based on total atoms = 7), corresponding to the ideal formula (Bi$_4$Te$_2$S$_2$).

Tsumoite was first discovered from the Tsumo mine (Shimazaki and Ozawa, 1978), but the occurrence of this mineral is rare in Japan. This mineral occurs as irregular grains of 1 to 20 \, \mu m in size within tetradymite (Figs. 2f and 4l). The empirical formula is (Pb_{0.01}Bi_{1.97}Sb_{0.01})_{2.01}Te_{1.91}S_{1.07} (based on total atoms = 2), corresponding to the ideal formula (Bi$_3$Te$_2$).

Rucklidgeite rarely occurs as small inclusions from 1 to 10 \, \mu m in size of tetradymite. The empirical formula is (Ag_{0.02}Cu_{0.03})_{2.01}Pb_{1.99}Bi_{1.95}Sb_{0.02} (based on total atoms = 9). The obtained unit cell parameters are $a = 23.8405(11)$, $b = 4.0479(2)$, $c = 19.0436 \, \AA$, $V = 1837.78 \, \AA^3$, $Z = 8$. Although this mineral is rich in Ag and Cu and its chemical composition is similar to neyite, the mineral is identified as cosalite by the unit cell parameters. The crystal structure of Ag and Cu rich cosalite has been reported by Topa and Makovicky (2010). The other occurs as irregular grains of 10 to 100 \, \mu m in size in association with tetradyomite and lillianite (Fig. 4h). The empirical formula is (Ag_{0.01}Cu_{0.01}Bi_{3.89}Sb_{0.02})_{1.91} (S_{4.97}Te_{1.00}Se_{0.01})_{1.02} (based on total atoms = 11), corresponding to the ideal formula (PbBiS$_3$).

**BI–SULFOSALT AND BI–TELLURIDE ASSEMBLAGES**

Bi–sulfosalts and Bi–tellurides from the Obari mine are divided into five assemblage types; (1) Type–A: wittichenite and aikinite–dominant assemblage, (2) Type–B: cuprobismutite group mineral–bearing assemblage, (3) Type–C: galenobismutite–bearing assemblage, (4) Type–D: Native Bi–bearing assemblage and (5) Type–E: Bi–telluride–dominant assemblage. Among these assemblages, Type–A is the most common and Type–B to Type–E are rare in the studied samples. (1) Type–A wittichenite and aikinite–dominant assemblage: This assemblage is composed of aikinite (Fig. 2a) and wittichenite (Fig. 2b), subordinate cosalite and tetradymite and trace amounts of miharaite. These Bi minerals are accompanied by chalcopyrite, bornite, tetrach-

(2) Type–B cuprobismutite group mineral–bearing assem-

Cuprobismutite group minerals from the Obari mine

185
This assemblage is composed of cuprobismutite group minerals (cuprobismutite, hodrušite, kupčíkite and paděraite), makovickyite, cupromakovickyite and aikinite-bismuthinite series (bismuthinite, krupkaite and aikinite) accompanied by emplectite and tetradymite (Fig. 4b), and trace amounts of nuffeldite and cannizzarite. Chalcopyrite and small amounts of molybdenite occur as other ore minerals in the assemblage.

(3) Type-C galenobismutite-bearing assemblage: This assemblage is composed of galenobismutite and cosalite, coexisting with small amounts of tetradymite and aikinite-bismuthinite series (friedrichite, hammarite lindströmite and pekoite) (Fig. 4e).

(4) Type-D native Bi-bearing assemblage: This assemblage is composed of aikinite, accompanied by minor native Bi, joséite-B, tetradymite, paderaite, aikinite-bismuthinite series (aikinite, hammarite and pekoite) and emplectite (Fig. 4k). Native Bi is very rare in the Obari mine.

(5) Type-E Bi-telluride-dominant assemblage: This assemblage is composed of tetradymite tellurobismuthite, tsumoite and rucklidgeite, in association with subordinate lilianite and cosalite (Fig. 2f). Chalcopyrite and galena sometimes occur in the assemblage.

DISCUSSION

Identification of cuprobismutite group minerals using minor components

Cuprobismutite group minerals from the Obari mine are distinguished by the amounts of the minor components in each mineral. When minor components of each cuprobismutite group mineral are plotted on a ternary diagram (Fig. 6), the amounts of minor components (Ag, Fe and Pb) in the minerals are consistent with the range shown by Topa et al. (2003a) and Topa and Makovicky (2006) except for cuprobismutite. Cuprobismutite from the mine is beyond the range, because it contains small amounts of Fe and is poor in Pb. However, this mineral is the most Ag-rich among the minerals and Fe-bearing cuprobismutite has been reported from other localities as discussed later. Therefore, the Fe-bearing cuprobismutite from the Obari mine is possibly identified as cuprobismutite. Consequently, it is considered that cuprobismutite group minerals from the mine are generally consistent with the characteristics of the minor components indicated by Topa et al. (2003a) and Topa and Makovicky (2006). In addition, the distribution of the minor components of hodrušite, kupčíkite and paděraite has been supported by X-ray intensity mapping (Fig. 5).

Characteristics of the chemical composition of cuprobismutite group minerals

In comparison with other localities and the ideal formula, cuprobismutite has a small amount of Fe. Fe-bearing cuprobismutite has been reported in previous studies. It is considered that the origin of the Fe is derived from Fe-bearing minerals as follows. The mineral from the Čierna Lehota, Slovakia contains small inclusions of pyrite (Pršek et al., 2005) and from the Szklarska Poręba quarry, Poland contains small lamellae of hodrušite which are intergrown with cuprobismutite (Pieczka and Gołąbiowska, 2012). Cuprobismutite from the Obari mine occurs as small grains and is adjacent to a Fe-bearing undetermined secondary Bi-mineral (Fe 2.6–6.0 wt%). The origin of Fe in cuprobismutite might be derived from the Fe-bearing secondary mineral.

Hodrušite from the mine is poor in Bi and rich in Cu, compared with other localities and the ideal formula (Fig. 7). A mineral similar to the one in this mine has been reported only from the Swartberg deposit, South Africa (Topa et al., 2003a). The crystal structure analysis of hodrušite from the Swartberg indicated that the Bi site is partly replaced by Cu (Topa et al., 2003a). It is considered that hodrušite from this mine is rich in Cu by the substitution of Bi → Cu as well.

Kupčíkite from the mine is poor in Bi and rich in Cu and Fe (Fig. 7), and contains trace amounts of Zn. This mineral has a similar composition to that reported from the Szklarska Poręba quarry (Pieczka and Gołąbiowska, 2012). The substitutions of Fe → Zn and Bi → Cu were
Cuproisbistrite group minerals from the Obari mine

Figure 7. (Bi + Sb + Ag) - (Cu + Fe) - Pb ternary diagram of cuproisbistrite group minerals from the Obari mine and other localities.

Many kinds of Bi–sulfosalts and Bi–tellurides such as the cuproisbistrite group minerals have been found in the Obari mine in this study. The total number of species of Bi minerals reaches more than 20. No deposit has been known with more Bi-minerals than the Obari mine in Japan. The Bi–sulfosalts of this mine are composed of Cu, Pb, Bi and S as major elements and Ag, Fe, Sb and Se as minor elements, though Ag-dominant Bi–sulfosalts such as matildite were not observed at this locality. Bi minerals from this mine are classified into four types; Cu–Bi–S minerals, aikinite–bismuthinite series (Cu–Pb–Bi–S minerals), Pb–Bi–S minerals and Bi–tellurides.

Emplectite and wittichenite, belonging to the Cu–Bi–S minerals have been reported from Cu deposits such as the Mihara (Sugaki et al., 1980) and the Imooka mines (Sugaki et al., 1975; Kitakaze et al., 2013) in which primary bornite occurs. In addition, miharaite and the eutectic intergrowths of bornite–chalcopyrite–wittichenite are known to occur at both mines as well as at the Obari mine. The occurrences of emplectite, wittichenite, miharaite and bornite reflect Cu–dominant mineralization. Pb–Bi–S minerals were not found in either mine, whereas they are common in the Obari mine.

The aikinite–bismuthinite series were discovered in the Cu–Bi–W–Co–As deposits such as the Sasanami mine, hosted in the Hoben granitic complex (Nakashima et al., 1981). Pb–Bi–S minerals such as canizzarite and cosalite were also found, but Cu–Bi–S minerals such as wittichenite and primary bornite were not present. The series from the deposits, as well as from the Obari mine, are associated with chalcopyrite and Pb-bearing minerals such as galena or Pb–Bi–S minerals. The occurrences of aikinite–bismuthinite series reflect Cu and Pb mineralization.

Pb–Bi–S minerals such as cosalite and lillianite have been reported in many type deposits in Japan such as the Cu–Bi–W–Co–As vein-type deposits hosted in the Hoben granitic complex (Nakashima et al., 1981), the Kamioka (Mariko et al., 1996) and the Nakatatsu (Mariko, 1981)
Zn–Pb skarn deposits, the Agenosawa Cu–Pb–Zn vein-type deposit in the green tuff region (Nedachi et al., 1973), the Suzuyama Sn vein-type deposit (Nedachi et al., 1990) and the Otome Mo–W vein-type deposit (Tsunoda and Shimizu, 1995). Like these deposits, the Pb–Bi–S minerals are associated with galena in the Obari mine. The occurrences of Pb–Bi–S minerals and galena reflect Pb mineralization.

Bi–tellurides such as tetradymite and tellurobismuthite have been reported in many types of Au–Bi–Te deposits such as the Tsumo (Sugaki et al., 1981b), the Sannotake (Kato and Sakurai, 1962) and the Shin-Kiura (Miyahisa, 1962) skarn deposit and the Oya vein-type deposit (Yamaoka, 1981). Bi–tellurides are associated characteristically with electrum in these deposits and Te is closely related with Au mineralization. In the Obari mine, electrum sometimes occurs with aikinite, tetradymite and chalcopyrite (Fig. 2e) and the Au grade of ores is Au 200 g/t (Shimazu et al., 1961), which is particularly high. The occurrences of Bi–tellurides reflect Au–Bi mineralization.

Ag–Bi–S and Ag–Pb–Bi–S minerals, such as maticlende and pavonite, have not been found in the Obari mine, but have been reported from the Sn–polymetallic deposits such as the Ikuno deposit (Shimizu and Kato, 1996; Shimizu et al., 1998) and the Suttsu deposit (Ono et al., 2004). Ag major minerals such as acanthite and Bi–selenides also have been found in Sn–polymetallic deposits. Exceptionally, makovickyite, belonging to pavonite homologous minerals, occurs in the Obari mine, but this mineral indicates Cu > Ag compared to the ideal formula (Izumino et al., 2013). In addition, Ag major minerals such as acanthite were not observed. On the other hand, benjaminite and pavonite, belonging to the pavonite homologous minerals, from the Ikuno mine indicate compositions close to the ideal formula and are Cu-free. Therefore, Ag–Bi–S minerals reflect Ag mineralization and it is considered that such minerals were not formed in the Obari mine due to the lack of Ag mineralization.

In conclusion, the Obari mine is characterized by the occurrences of Cu–Bi–sulfosalts, reflecting the Cu–dominant mineralization. Moreover, since this mineralization is associated with small amounts of Au–Te and Pb, many kinds of Bi–sulfosalts and Bi–tellurides were formed, compared to those of other deposits in Japan. On the other hand, Ag–Bi–S minerals were not formed because of a lack of Ag in the mineralization unlike Sn–polymetallic type deposits.

ACKNOWLEDGMENTS

Special thanks are due to Mr. Jun-Ichi Igarashi for providing the ore specimens from the Obari mine and Prof. Richard W. Jordan of Yamagata University for improving the English manuscript. We thank Dr. Ritsuro Miyawaki, an anonymous reviewer and Associated Editor Dr. T. Mikouchi for their constructive and editorial comments.

SUPPLEMENTARY MATERIAL

Color version of Figures 2 is available online from http://japanlinkcenter.org/DN/JST/JSTAGE/jmps/140129.

REFERENCES


