Comparison of the cytotoxic effects and smear layer removing capacity of oxidative potential water, NaOCl and EDTA

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Abstract: This study compared the smear layer removing capability and cytotoxicity of NaOCl, EDTA and Oxidative Potential Water (OPW). Fifteen extracted single-rooted human upper incisors were examined in three groups. The root canals were enlarged to the apical foramen with K files to size #60 and irrigated with: (a) NaOCl followed by OPW, (b) OPW during and after instrumentation and (c) NaOCl followed by EDTA and NaOCl. The effect of these irrigants on the smear layer was evaluated using a scanning electron microscope. In vitro cytotoxicity of these irrigants was examined by MTT colorimetric assay. We found that the combination of NaOCl and OPW as well as the application of OPW alone, failed to remove the smear layer from the apical third, whereas the EDTA and NaOCl combination achieved complete removal. OPW, when used during and after instrumentation, removed the smear layer in the middle third more effectively than NaOCl followed by OPW. EDTA exerted more cytotoxic effects at all concentrations tested when compared with OPW and NaOCl. In conclusion: (a) OPW was less cytotoxic than other irrigants but did not effectively remove the smear layer, (b) treatment with EDTA followed by NaOCl efficiently removed of the smear layer, but their cytotoxicity should be considered during endodontic therapy. (J. Oral Sci. 43, 233-238, 2001)

Key words: irrigation solution; smear layer; cytotoxicity.

Introduction

Irrigation solutions have been used in endodontics for removing necrotic tissues, smear layer, bacteria and debris from the root canal (1,2). Although numerous studies have been done in vitro, the relative effectiveness of different irrigants has not been clearly demonstrated in clinical usage (3).

Sodium hypochlorite (NaOCl) is widely used as an irrigation solution because of its tissue dissolving, lubricant and also antibacterial properties (1-5). In order to obtain acceptable bactericidal and solvent effects, 5.25% NaOCl solution has been recommended (2). However, effective concentrations of this solution are reported to be cytotoxic (6,7). The cytotoxicity of NaOCl is reduced at lower concentrations, but dilution impairs its irrigant properties (2,5,8). It is generally agreed that there is no single solution which has the ability to remove both organic and inorganic components of the smear layer. Therefore, it was recommended that the most effective procedure to adequately clean the root canal system is to irrigate the canals with ethylenediaminetetraacetic acid (EDTA) followed by NaOCl (9-11).

Recently, Hata et al. (12) studied a new antibacterial and antiviral agent for root canal irrigation, Oxidative Potential Water (OPW), which was developed in Japan for disinfecting purposes. In their study, OPW was reported to be effective in removing the smear layer. With the absence of any toxicity, and with its antimicrobial properties and smear layer removal capability, OPW seems to be an...
acceptable endodontic irrigation solution.

The purpose of this study was to investigate the potential of OPW as an irrigation solution for endodontic treatment. The questions to be tested was whether the OPW was an acceptable endodontic irrigation solution regarding cytotoxicity and efficacy in smear layer removal, and whether it can be an alternative to NaOCl and EDTA.

**Materials and Methods**

**Scanning electron microscopic (SEM) examination**

Fifteen extracted single-rooted human upper incisors were used in this study. Following extraction, the teeth were stored in a physiological saline solution at 37°C for 48 h. The access cavities were then prepared and the working length was established. The teeth were randomly divided into three groups and the root canals were enlarged to the apical foramen with K files to size #60 by using the step-back technique, producing a standard flare by the insertion of #3-5 Gates Glidden drills. The specimens were irrigated as follows: Group 1: 2 ml of 5% NaOCl (ACE, Procter & Gamble, Istanbul, Turkey) was used between each instrumentation size and the final irrigation was accomplished with 10 ml of OPW (NDX-250KH, Nihon Aqua Co. Ltd., Kyoto, Japan) for 2 min.

Group 2: 2 ml of OPW was used between each instrumentation, and 10 ml of OPW was used for 2 min as the final flush.

Group 3: 2 ml of 5% NaOCl was used between each instrumentation and then the canals were flushed with 10 ml of 17% EDTA (Sigma Chemical Co., St. Louis, MO) for 2 min followed by 10 ml of 5% NaOCl.

The crowns of the teeth were removed with a high-speed handpiece with a water spray at the cementoenamel junction. All the roots were then sectioned longitudinally. These root samples were fixed with 2.5% phosphate-buffered gluteraldehyde at 4°C for 9 days. After fixation, samples were washed with phosphate buffer and dehydrated with increasing concentrations of acetone (25, 50, 75, 100%). Dehydration was completed with a critical point dryer. Samples were coated with 200Å-thick gold using a gold-coating apparatus and examined for SEM (JEOL-SEM 6400) evaluation.

**Cytotoxicity test**

**Cell Cultures.** L929 mouse fibroblasts (ATCC) were cultured in 25 cm² culture flasks containing RPMI-1640 supplemented with 10% fetal bovine serum, 2 mM L-glutamine (Sigma Chemical Co., St. Louis, MO), 100 μg/ml of streptomycin, 100 μg/ml of penicillin (Sebak, Biologische Fordchungs GmbH, Germany) at 37°C in a humidified incubator under an ambient air pressure atmosphere containing 5% CO₂. Confluent cell monolayers were trypsinized; after the fourth passage, cells were used in the cytotoxicity experiments as described below. The irrigation solutions were filtered through a 0.2 μm filter (Minisart NML, Sartorius GmbH, Germany). Various dilutions (final dilution: 1/2, 1/4, 1/8 and 1/16) were prepared in order to use in the cytotoxicity assay.

**MTT Assay.** A simple colorimetric assay developed by Mosmann (13) as a test for cell proliferation and survival was adapted for the measurement of cytotoxicity. This assay involves the ability of viable cells to convert a soluble tetrazolium salt, MTT [3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide], into a blue formazan endproduct by mitochondrial dehydrogenase enzymes. The blue color reaction is used as a measure of the cell viability. The assay is performed in 96-well plates and results are read on a multiwell spectrophotometer. A modified method of the colorimetric MTT assay was used in this study (14). For this purpose, 2×10⁴ cells in 50 μl of culture medium were seeded in flat-bottomed microplates (Costar, Cambridge, MA) and 50 μl of sterile irrigation solution was added to each well. Cells in 50 μl of culture medium alone served as a control for the cell viability. The assay was run in quadruplicate so that control and dilution values were obtained as the mean values of four identical wells.

After a 24 h incubation at 37°C under a humidified, 5% CO₂-containing air atmosphere, 25 μl of 5 mg/ml MTT (Sigma Chemical Co., St. Louis, MO) in saline was added into each well and incubated for a further 3 h at 37°C. A solubilization buffer consisting of 23% sodium deoxycholate sulphate (SDS; Sigma Chemical Co., St. Louis, MO) in 50% N,N dimethyformamide (pH 4.7) was used to dissolve the formazan precipitate. For this purpose, 80 μl of solubilization buffer was added into each well and plates were then incubated overnight at 37°C. The optical densities (OD) of the plates were read by a microplate spectrophotometer (Coulter, Electronics, Inc., Hialeah, FL) at 570 nm.

**Analysis of cytotoxicity.** Analysis of the potential cytotoxicity of the irrigation solutions under investigation was conducted as follows. First, the optical densitometric readings of each of the four wells were determined and the arithmetic mean ± standard error of these four recordings were calculated. Second, an arbitrary unit of 100 was assigned to the optical densitometric reading obtained from the control wells. The optical densitometric readings obtained from the control wells and subsequent dilutions
were then converted to percentages. Then, control, 1/2, 1/4, 1/8 and 1/16 dilutions were plotted as the abscissa of a linear regression graph with corresponding percentages of optical densitometric values as the ordinates. By using Instat Statistical Software, a linear regression line was fitted onto the converted values and the dilution on the abscissa that produced 50% cytotoxicity (Cytotoxic Dilution 50%, abbreviated CD50) was determined by interpolation on the regression line. These CD50 values were used as an expression of the cytotoxicity potentials of the substances and they were used for comparison in accompaniment with corresponding correlation coefficient values ($r^2$). The effect of irrigation solutions was evaluated by comparing the mean optical density (OD) value of treated cells to the mean OD of untreated control wells.

**Results**

**SEM examination**

Group 1: Specimens irrigated with 5% NaOCl, followed by OPW, showed a heavy smear layer in the apical third. Dentinal tubule openings were obscured by extensive debris (Fig. 1). The middle third of these specimens appeared to be relatively clean. Tubule outlines were hardly visible and all of the tubular apertures were plugged with the smear material (Fig. 2).

Group 2: Specimens irrigated during and after instrumentation with OPW, displayed similar findings with group 1 in the apical third. Smear plugs covered the apertures of...
the dentinal tubules (Fig. 3). The smear layer was almost effectively removed, but a slight amount of debris was observed that obscured tubule orifices in the middle third (Fig. 4). In longitudinal sections, a slight penetration of plugs into the tubules was clearly observed in the same area (Fig. 5).

Group 3: Irrigation of the specimens with NaOCl followed by EDTA and NaOCl resulted in complete removal of the smear layer in the apical and middle third. In these specimens, enlargement of the dentinal tubule openings was observed (Fig. 6).

Cytotoxicity assays with solutions
The cytotoxicity of irrigation solutions OPW, EDTA and NaOCl at different concentrations was assessed in L929 cells using the MTT assay. The results of the cytotoxicity experiments, which illustrate the percentage of OD values of irrigation solution-treated cells to the OD value of control cells, are presented in Table 1 and cytotoxic potentials of these irrigation solutions are given in Table 2. Although the diluted concentrations of OPW and NaOCl did not exert toxic effects, we observed 50% cell death in lower dilutions. CD50 values for OPW and NaOCl are 1/2 and 1/3.7, respectively. EDTA was found to be very toxic at all dilutions tested.

Discussion
Successful root canal treatment depends on proper chemical and mechanical debridement of the root canal system. Since no single agent has been found to be sufficient for this purpose, various combination of irrigants have been studied (10-12,15).

In this study, we evaluated three different irrigation solutions (OPW, NaOCl and EDTA) in combination. OPW, which is strongly acidic with a pH of 2.5, is produced by electrolysis of tap water containing a small quantity of NaCl. We found that instrumentation of the root canal with 5% NaOCl followed by OPW and OPW-OPW treatment failed to remove the smear layer in the apical third and left additional extensive debris at the dentinal openings. On the other hand, we also found that OPW, when used during and after instrumentation, removed the smear layer in the middle third more effectively than NaOCl followed by OPW (but the tubule openings were still clogged). However, Hata et al. (12) reported that OPW, when used during and after root canal instrumentation, resulted in the smear layer on the root canal surface not being observed and the tubular packing phenomenon not being observed in the tubules. This discordance may be due to the differences in technical manipulations. Hata et al. (12) also reported that OPW in combination with EDTA cleans the root canal surfaces retaining no smear layer. Our study showed that NaOCl followed by EDTA and NaOCl irrigation achieves complete smear layer removal on the apical and middle third of root canal, although erosive effects were observed at the tubular openings. EDTA is a strong chelator, exerting a demineralizing effect reacting with calcium ions in the hydroxyapatite crystals of dentin. However, the smear layer removing capability should be considered with its toxic and erosive effects. Ideally, an irrigation solution is expected to remove the smear layer with as low of a toxicity as possible. Normally, as the concentration of a solution is increased, unwanted cytotoxic effects are produced. OPW is accepted to be safe for patients to hold in the oral cavity, due to its ability to lose the high oxidation-
reduction potential and low pH upon reacting to light and/or organic substances (12). Our results show that OPW has lower cytotoxicity than NaOCl and EDTA. In this study, we found that the cytotoxicity of EDTA was remarkable at any dilution as evaluated by MTT assay (16,17). These findings are in agreement with Koulaouzidou et al. (7) who reported that at 17%, 15% and 1%, EDTA demonstrated severe cytotoxicity in vitro. According to Segura et al. (18), the apical extrusion of EDTA not only causes a decalcifican action on periapical bone, but it also may have effects on neuroimmune regulation, even when the concentration of EDTA is very low. They also reported that leakage of EDTA to periapical tissues during root canal preparation may inhibit macrophage function and reduce periapical inflammatory reactions (19).

Our findings indicate that OPW has the lowest cytotoxicity, but could not completely remove the smear layer. For this reason, it can best be utilized in combination with a chelator substance which has better smear layer removal characteristics. We believe that OPW may be considered as an alternative irrigation solution, however, finding the best combination needs further study. In addition, antimicrobial and tissue dissolution properties of OPW must be investigated.

Table 1 The percentage of OD values of irrigant-treated cells to the control cells

<table>
<thead>
<tr>
<th></th>
<th>Control</th>
<th>1/16</th>
<th>1/8</th>
<th>1/4</th>
<th>1/2</th>
</tr>
</thead>
<tbody>
<tr>
<td>OPW</td>
<td>100</td>
<td>97.72</td>
<td>88.40</td>
<td>75.81</td>
<td>48.42</td>
</tr>
<tr>
<td>NaOCl</td>
<td>100</td>
<td>54.27</td>
<td>51.16</td>
<td>44.59</td>
<td>37.45</td>
</tr>
<tr>
<td>EDTA</td>
<td>100</td>
<td>30.59</td>
<td>29.34</td>
<td>26.95</td>
<td>24.89</td>
</tr>
</tbody>
</table>

Table 2 Cytotoxic potentials of irrigation solutions on L929 cells

<table>
<thead>
<tr>
<th></th>
<th>CD50</th>
<th>r²</th>
</tr>
</thead>
<tbody>
<tr>
<td>OPW</td>
<td>1/2</td>
<td>0.9944</td>
</tr>
<tr>
<td>NaOCl</td>
<td>1/3.7</td>
<td>0.5351</td>
</tr>
<tr>
<td>EDTA</td>
<td>1/9.2</td>
<td>0.3425</td>
</tr>
</tbody>
</table>

D₅₀ stands for cytotoxic dilution 50; for description see Materials and Methods section.
r²: correlation coefficient.

References
15. Berg, M.S., Jacobsen, E.L., BeGole, E.A. and