Two-Dimensional Time-Resolved X-ray Diffraction Study of Directional Solidification in Steels

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In situ characterization of directional solidification process during welding was carried out using the time resolved X-ray diffraction technique utilizing intense synchrotron radiation. Then, behaviour of dendrites in steels under welding conditions of a practical manufacturing process were investigated using the TRXRD method for *in-situ* weld observation with the uniquely-sensitive two-dimensional pixel detector. Consequently, the crystal growth during the rapid cooling was caught in detail and employed a systematic peak profile analysis in order to acquire the essential information for controlling the weld microstructure. Our results would suggest the microstructure formation process of low alloy in directional solidification during rapid cooling. Simultaneously, we discuss the possibility of detecting the nucleation.
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1. Introduction

High intensity heat sources used for fusion welding create steep thermal gradients, which produce phase and microstructural changes in materials as they are rapidly heated and cooled to and from their melting points. The rapid thermal cycling that takes place induces melting and solidification in those parts of the weld where the liquidus temperature has been exceeded, as well as solid state phase transformation on both heating and cooling during welding. When steels are welded, the optimized base metal properties are altered by the localized weld thermal cycle. The result is the creation of non-equilibrium microstructures in the weld. These microstructures are significantly different in both appearance and properties from those found in the base metal. Such non-equilibrium phases can compromise the integrity of the weld joint, making it important to understand how the welding conditions lead to their creation. However, it was difficult to observe the microstructural changes of weld metals dynamically, under rapid cooling over several hundred Kelvin per second. So the solidification morphology has been observed by the Scanning Electron Microscope (SEM), using the liquid-tin quenching method in high alloys that have some influence on solid-transformation.¹⁻³⁾ Since late nineties, Elmer *et al.*⁴⁻⁶⁾ has used intense synchrotron radiation and has directly revealed the phase mapping around the heat source, during pure titanium welding. After that, Babu *et al.* investigated the primary weld solidification in the Fe-C-Al-Mn steel weld⁷⁻⁹⁾ using the Time-Resolved X-Ray Diffraction (TRXRD) technique. Their remarkable results were devoted to the heat affected zone (HAZ). In contrast, several researchers observed the real-time growth of dendrites in alloys using synchrotron microradiograph in the real space.¹⁰⁻¹²⁾ However, it is impossible to obtain the quantitative information such as phase transformation and preferred orientation in the real space.

Our interest is details of the weld solidification phenomena

in the directional solidification process under rapid cooling. With the availability of intense X-ray beams from synchrotron storage rings, it is now possible to make a direct observation of phase transformation and microstructural evolution *in situ* and in real time as a function of welding time. In our previous work, a numerical model to predict the microstructure of austenitic stainless steel weld metal was proposed, and spatially resolved X-ray diffraction using synchrotron radiation was carried out for weld metals of high alloys, quenched in liquid tin, to verify the validity of the numerical mode. The crystallization temperatures predicted by the numerical model for secondary γ and δ phases in weld metals agreed with experimental data.¹³⁾ Further, combination of analyzing method: the *in-situ* phase identification system, morphological observation by high-temperature laser scanning confocal microscopy and observation of microstructure at room temperature by optical microscope, scanning electron microscope and micro diffraction-system, is suggested to analyze the phase transformation during welding process. Consequently, phase transformation process of hypereutectoid carbon steel, during welding was analyzed as an example of a combination observation.¹⁴⁻¹⁶⁾ Furthermore, we carried out *in-situ* observation for weld solidification in stainless steels using time-resolved X-ray diffraction. We assumed the rotation of dendrites from the discontinuous diffraction pattern recorded by the imaging plate along one-direction of the reciprocal space.¹⁷⁻²¹⁾ Then, the nucleus of dendrites was observed in the mist-like diffraction pattern of the initiate stage in high alloy. Although these experiments have had some success in the observation of a solidification process, the influence of the crystal preferred orientation by eutectic growth was not avoided. In the present work, behavior of dendrites in steels under welding conditions of a practical manufacturing process were investigated using the TRXRD method for *in-situ* weld observation with the uniquely-sensitive two-dimensional pixel detector.²²⁻²⁴⁾ The crystallographic information of the δ and γ phases in a

Fig. 1 The phase transformation sequence calculated from equilibrium thermodynamic relationships using Thermocalc software version Q with the iron based alloy database. The thermodynamic calculations indicate that the equilibrium microstructure of these alloys after solidification consists of a mixture of ferrite and Fe_3C carbide phases. The phase transformation sequence in Fe-0.02%C and Fe-0.88%C is shown by a solid line (a) and a break line (b).

directional solidification during rapid cooling was obtained for the first time. Our results would suggest the microstructure formation process of low alloy in directional solidification during rapid cooling. Simultaneously, we discuss the possibility of detecting the nucleation.

2. Experiments

The specimens were prepared for a low alloy with the basic concentration of Fe-0.02%C and Fe-0.88%C. The carbon content of steels play an important role in the microstructural evolution of the weld. The phase transformation sequence was calculated from equilibrium thermodynamic relationships using Thermocalc software version Q with the iron based alloy database as shown in Fig. 1. The thermodynamic calculations indicate that the equilibrium microstructure of these alloys after solidification consists of a mixture of ferrite and Fe_3C carbide phases. During cooling, the ferrite (δ phase) first appears at 1500°C in Fe-0.02%C. This microstructure begins to transform to austenite (γ phase) when the temperature reached 1450°C . It begins to transform to the ferrite (α phase) when the temperature reached 800°C . Complete transformation to ferrite occurs when the temperature reached 730°C . Then, these ferrite and Fe_3C carbide phases remain stable. On the other hand, in Fe-0.88%C, this microstructure begins to transform to austenite (γ phase) when the temperature reached 1500°C . Complete transformation to austenite occurs when the temperature reached 1450°C . It begins to transform to the ferrite (α phase) when the temperature reached 730°C , and these ferrite and Fe_3C carbide phases remain stable. However, kinetic limitations may alter the predicted phase transformation start and

completion temperatures, and may produce non-equilibrium phases.

In the TRXRD, the time series transition of the diffraction patterns in the solidification process of welding was measured. The XRD experiments were performed on the BL46XU beam line with the undulator at the SPring-8 with the approval of the Japan Synchrotron Radiation Research Institute (Hyogo, Japan).²⁵⁾ The experimental set-up of original TRXRD technique for welding is summarized schematically in Fig. 2. The time resolved diffraction patterns in the solidification process was measured while a torch moves vertically to an X-ray incident beam at the speed of 1.0 mm/s by an automatic stage. The incident angle (α) of X-ray beam was fixed at 10° . Further, the welding assembly was tilted as the arc is vertically to the surface of sample. Then, the 9-mm-wide fusion zone was formed on the surface of the steel plate. The photon energy of 18 KeV was chosen to maximize the intensity and the number of peaks into the 2θ window of the detector. The wavelength calibrated by the Rietveld refinement of the LaB_6 standard specimen is 0.0689 nm .^{26,27)} Adjustment of the specimen-to-detector distance (camera length), is best made by distances between five direct spots with five degree intervals (r). The camera length (D) was decided very simply from the relation $D \cdot \tan 2\theta = r$. The slit size is $W0.1\text{ mm} \times L0.5\text{ mm}$. Then, radiation area is about $W0.1\text{ mm} \times L2.8\text{ mm}$ in the perpendicular to the direction of dendrite growth for incident angle (α) of 10° . It is small for 9–10 mm in the width of the weld bead. Then, the penetration depth of X-ray is guessed to be about $10\text{ }\mu\text{m}$. The dispersive angle of X-ray is guessed to be about 20 arcsec from the reflectivity of the silicon wafer.

The TRXRD data was collected at a single location during welding. The weld metals have the preferred orientation by $\langle 100 \rangle$ eutectic along the growth direction and the weak broad pattern. Therefore, it is difficult to simultaneously record the first and second phases. Consequently, diffraction patterns collected with a uniquely-sensitive two-dimensional pixel detector that was developed by SPring-8 detector team in collaboration with Paul Scherrer Institute (PSI) in Switzerland. The counters can be read out digitally with a frequency of 10 MHz . It leads the maximum frame rate to 30 Hz with the readout time of 6.7 ms . This detector was mounted on the arm of four-axis diffractometer at a distance of approximately 286 mm behind the weld to cover a 2θ range from 19 to 51 deg . This 2θ range was optimized to contain a total of five diffraction peaks, three from BCC phase ($\alpha\text{-Fe}$ or $\delta\text{-Fe}$) and two from the FCC phase ($\gamma\text{-Fe}$). That is, q_x and q_y of reciprocal space coordinates ($= 4\pi \sin \theta / \lambda$) become $30\text{--}79$ and $-12\text{--}+12$ respectively. Therefore, the lattice spacing of $0.08\text{--}0.2\text{ nm}$ can be measured.

The cooling rate, i.e., solidification speed, is controlled by the arc power and the scan speed of the torch. The power was maintained at a constant 1.5 kW (150 A , 10 V) for all of the welds, and high pure argon was used as a shielding gas. The heating temperature on the back of the specimen was prevented by a water-cooled copper plate and a ceramic insulation material. TRXRD experiments provide information about phase transformation as a function of weld time, but do not directly provide information about weld temperatures. In order to relate weld time to weld temperature, the

Fig. 2 The experimental set-up of original TRXRD technique for welding. The photon energy of 18 KeV was chosen to maximize the intensity and the number of peaks into the 2θ window of the detector. The incident angle (α) of X-ray beam was fixed at 10° . The time resolved diffraction patterns in the solidification process was measured while a torch moves vertically to an X-ray incident beam at the speed of 1 mm/s by the automatic stage. The arc-power was maintained constantly at 1.5 kW (150 A, 10 V) for all of the welds. In order to have a preferred orientation, we utilized a uniquely-sensitive two-dimensional pixel detector.

temperature of the weld metal was measured continuously by the thermocouple of type R.¹⁷⁾ We utilized one of the cases that the thermocouple did not react with molten steels and where it contacted enough with molten steels.

Our technique can capture the change of the dendrite microstructure that grows up after weld melting. The direction of incident beam is very important as each dendrites are considered like single crystals. That is, dendrites that grow in the parallel to the arc movement would contribute to the diffraction. In contrast, dendrites that rotate by axis of plane normal of sample would not contribute to the diffraction. Therefore, dendrites that grow up epitaxially from surface of Heat Affected Zone (HAZ) heated just under melting point would not contribute to the diffraction. Meanwhile, though J. Wong *et al.* perform the same kind of experiment,²⁸⁾ their techniques would catch microstructural change of the spot solidifications by on/off of heat source. Therefore, the processes of the dendrite growth would be controlled by epitaxial growth from HAZ. It is not directional solidification process for welding cycle.

3. Results and Discussion

Figure 3 shows some information of Laue patterns about phases that exist at all times during the weld cycle. For Fe-0.02%C, Laue patterns are indicated at approximately (a) 1500°C, (b) 1450°C, (c) 600°C, (d) 500°C and (e) 400°C and for Fe-0.88%C at approximately (f) 1500°C, (g) 1300°C, (h) 700°C, (i) 600°C and (j) 400°C during welding. The x-y axis is corresponded to that in Fig. 2. The origin of the reciprocal lattice is on the left hand side parallel to x axis. Therefore, the right hand side corresponds to high 2θ value. In Fe-0.02%C, the diffraction pattern of the $\delta 220$ first appears at approximately 1500°C as shown in Fig. 3(a). The diffraction pattern of the δ phase begins to transform to the secondary γ phase at 1450°C as shown in Fig. 3(b). At 600°C, the diffraction pattern of α phase appears while the diffraction intensity of γ

Fig. 3 Change of the diffraction pattern for the high alloy (a-d) and low alloy (e-h) during welding. (a) $\sim 1450^\circ\text{C}$, (b) $\sim 1400^\circ\text{C}$, (c) $\sim 1350^\circ\text{C}$, (d) $\sim 1000^\circ\text{C}$, (e) $\sim 1500^\circ\text{C}$, (f) $\sim 1450^\circ\text{C}$, (g) $\sim 600^\circ\text{C}$ and (h) $\sim 400^\circ\text{C}$. The x-y axis corresponded to that in Fig. 2.

phase decreases as shown in Fig. 3(c) suggesting $\gamma \rightarrow \alpha$ solid-state transformation. Then, a somewhat finer grain size increases the number of Laue spots and the suggestion of Laue rings begins to appear. A still finer grain size produces the smooth, continuous Laue ring as shown in Fig. 3(d). That is, the α phase continues to increase in amount as the $\gamma \rightarrow \alpha$ solid-state transformations proceed. Finally, the diffraction pattern of the residual γ and α phases mainly coexist as shown in Fig. 3(e). The solid-transformation to the pearlitic microstructure which consists of ferrite and cementite during cooling is well known. Therefore, it is expected that Fe_3C phases would simultaneously appear with α phase. However, peaks of Fe_3C phases could not be observed as the diffraction intensity were weak in angular range. Consequently, it is considered that formation of the continuous Laue-ring is referred to miniaturization of microstructure due to generation of the α and Fe_3C phases in the final stage of solidification. Further, it is expected that the α phase would transform to the γ phase by having a preferred orientation. The main orientation seems to be $\gamma 220 // \alpha 211$ by the angular relationship of the diffraction spots of γ and α phases. However, the generation of Body-Center-Tetragonal (BCT) lattice is known by $\gamma 220 // \alpha 200$ in the solid transformation.²⁹⁾ Therefore, it is considered that dendrites that have grown up from the inside of the sample to the surface would influence on the diffraction. Further, the peak width on the projected images gradually increased. This broadening of the tail by increase of intensity means increase of phase with preferred orientation.

In Fe-0.88%C shown on the right hand side in Fig. 3, the

diffraction pattern of γ phases first appears at about 1500°C as shown in Fig. 3(f). The diffraction intensity of the γ phase gradually increases as shown in Fig. 3(g). At 700°C, the diffraction spots of the γ phase extend in y -direction, while the diffraction pattern of secondary α phase appears as shown in Fig. 3(h). These diffraction intensities of the α phase gradually increase while the diffraction patterns of γ phase become weak as shown in Fig. 3(i). Finally, the diffraction pattern of the residual γ and α phases mainly coexist as shown in Fig. 3(j). The comparatively random microstructure is formed as shown by the continuous ring pattern in Fig. 3(j). It is considered that this phenomenon is referred to miniaturization of microstructure due to the solid-transformation, i.e. generation of the α and Fe_3C phases, in the final stage of solidification. When the grain size is further reduced, the Laue spots merge into a general background and only Debye rings are visible. Therefore, it is expected that grains in Fe-0.88%C distribute finely and randomly than those in Fe-0.02%C. Further, the orientation relationship differs from that of Fe-0.02%C, and not only $\gamma 220 // \alpha 211$ but also $\gamma 220 // \alpha 200$ is observed. The relationship of $\gamma 220 // \alpha 200$ corresponds to Kurdjumov-Sachs (K-S) relationship.^{30,31} That is, it is considered that the original relationship appeared by the progress of minimization of microstructure.

The TRXRD results presented above provide information about the crystallographic parameters of phases as a function of weld cooling temperature. There is a wealth of information about the lattice parameter of each phase and how they change with welding time, plus other information such as the width, intensity, and areas of each of the diffraction peaks during the welding cycle. The change of the diffraction pattern of $\gamma 220$ in Fe-0.88%C during rapid cooling was quantitatively plotted in Fig. 4(a). The displacement of the lattice parameter and the Full-Width of Half-Maximum (FWHM) along the x -direction were shown on the right axis and left axis as a function of weld temperature, respectively. The lattice constant is converted into uniform strain. Convergent of the lattice spacing to the equilibrium value means the increase of crystallinity. The lattice spacing can be determined from the 2θ positions of the diffraction peaks of $\gamma 220$ using Bragg's law and normalized energy of the incident beam. In contrast, the FWHM is converted into the size of crystallites and non-uniform strains. If a material expands or contracts as a function of temperature without undergoing a phase transition, the position of the reflections in the diffraction pattern change. If the crystallite is given a uniform tensile strain at right angles to the diffraction planes, their spacing becomes larger than equilibrium spacing, and the corresponding diffraction line shifts to lower angles but does not otherwise change. The grain is bent and the strain is non-uniform, on the tension side the Bragg plane spacing exceeds equilibrium spacing, on compression it is less than equilibrium spacing, and somewhere in between it the equilibrium spacing. Thus, a crystallite can be thought of as composed of a number of small regions each of which the plane spacing is substantially constant but different from the spacing in the adjoining regions. Consequently, it should be noted that the crystal growth of dendrites mainly consists of two stages. First, the FWHM rapidly becomes narrow up to

Fig. 4 The change of the diffraction pattern of $\gamma 220$ in Fe-0.88%C during rapid cooling.

about 800°C, suggesting the increase of crystallites size. Second, the lattice constant rapidly shift to the low angle from about 800°C, suggesting the improvement of crystallinity and the solidification shrinkage. The crystallinity improves from about 900°C after the crystallites size increase rapidly. The lattice constant decreases about 0.005 nm between 977 and 415°C. In a word, the fraction of shrinkage ratio per Kelvin in the rapid cooling process is $1.9 \times 10^{-5}/\text{K}$. It corresponds to the thermal expansion coefficient $2 \times 10^{-5}/\text{K}$ of general low carbon steel.³²

Next, Figure 5(a) shows the angular displacement of the $\gamma 220$ diffraction spot on the Laue-ring. Fig. 5(b) shows the phase fraction calculated by *thermocalc* software. The circles in the Fig. 5(a) mean a displacement angle based on 1400°C. This displacement means a geometrical tilt in the normal to the direction of dendrite growth. This is different from the rotation that centers on the direction of dendrite growth suggested by high-alloy steels.¹⁷ Considering geometry, the diffraction spots would blink on the two-dimensional diffraction pattern by rotation of the dendrite. However, the diffraction spot of $\gamma 220$ in the low-alloy is caught as a continuous change from the mist-like pattern to the sharp spot. Further, the crystallites tilt rapidly more than 1300°C. Compared to the calculated phase fraction, this displacement may be synchronized to decrease liquid as shown by solid line in Fig. 5(b). That is, it would be behavior of the crystallites that is mobile in the liquid. Next, the crystallites stop tilting at 800°C and tilt greatly until 400°C. The crystallites stop tilting again at 400°C. The dendrites tilt in total by approximately one degree. A solid line in Fig. 5(a) is the differential of a break line, showing the tilt angle of crystallites for approximately 1°C. The maximum absolute value is in the vicinity of about 650°C. It is thought that the refinement of the microstructure influences formation of the α and Fe_3C phases around 650°C. A steep increase in the α phase (BCC) is shown by the break line in the equilibrium diagram calculation in Fig. 5(b). However, the microstructure continuously tilts in non-equilibrium rapid cooling because the α phase increases continuously in our experiment. That is, there would be possibilities of precursor of the phase transformation, and fluctuation of the eutectic point. Moreover, it seems that the microstructure formation due to the phase transformation was completed at 400°C since the

Fig. 5 Angular displacement of the γ_{220} and phase fraction. (a) the angular displacement of the γ_{220} diffraction spot on the Laue-ring. The circles mean a displacement angle based on 1400°C. A solid line is a differential of a break line, showing the tilt angle of crystallites per degree C. (b) the phase fraction calculated by *thermocalc* software. Comparing with the calculated phase fraction, this displacement may be synchronized to the decrease of liquid as shown by solid line.

crystallites stop tilting again. Thus, there are two tilting zones, one is due to crystallites in the liquid and the other is due to formation of a microstructure.

The change of diffraction pattern was observed in detail in order to reveal the growth process of dendrites. Figure 6 shows images of γ_{220} diffraction pattern in the Fe-0.88%C in the initial stage. Diffraction patterns (a, b, c and d) of γ_{220} corresponds to 1440, 1350, 920 and 420°C, respectively. The Fig. 6(a) indicates the weak broad pattern which we could observe at first. The slight sharpening of the peak profile at 1350°C is clearly evident in the pixel detector record, and so is the doublet resolution which occurs at 920°C. But note that the pixel detector can not observe spotty diffraction patterns. The pixel detector record would immediately suggest that the specimen at 420°C had high crystallinity. Thus, the diffraction pattern changes remarkably from the isotropic weak broad pattern like a mist to the sharp spot along with the temperature drop. The change of these profiles were shown quantitatively by a high temperature range in Fig. 4(a). In our previous work, it was expected that the mist-like pattern would be correlated with the nucleus of dendrites in the high alloy.¹⁷⁾ That is, the phenomena in the vicinity of the critical nucleus may be observed on average as weak broad patterns that are scattered by the wide distribution of lattice spacing and atomic distance. Dendrites behaviour in the low alloy would be also similarly expected in the high alloy. However, the behaviour of dendrite growth is different from each other.

The section along the x -direction in Fig. 6(a) was quan-

Fig. 6 Images of γ_{220} diffraction pattern in the Fe-0.88%C in initial stage. Diffraction patterns (a, b, c and d) of γ_{220} corresponds to 1440, 1350, 920 and 420°C, respectively. The section along the x -direction in Fig. 6(a) was quantified in order to observe the behaviour of crystallite in the initial stage. The results of the peak separation are shown in Fig. 6(e). The crystallite size can be determined easily by the Scherrer method through individual peak profile analysis. Then the integral breadth is given by the integrated intensity divided by the maximum intensity. The experimental data are shown as the break line. The experimental data was smoothed along the x -direction after removal of background. While, solid lines represent separated overlapping peaks.

tified in order to observe the behaviour of crystallite in the initial stage. The results of the peak separation is shown in Fig. 6(e). The crystallite size can be determined easily by the Scherrer method through individual peak profile analysis.³³⁾ Integral breadth is given by the integrated intensity divided by the maximum intensity, i.e., it is the width of a rectangle having the same area and height as the observed line. The experimental data is shown by the break line. The experimental data was smoothed along the x -direction. While, solid lines represent overlapping peaks separated by Gaussian fitting. It corresponds to the temperature drop of approximately 30°C in the vicinity of 1400°C since the exposure time is 0.15 seconds. The peaks of different angles appear by fitting on two overlapping peaks since peak profile has a shoulder. Therefore, two-kinds of components are expected for two peaks between approximately 30°C. The

individual crystallite of such a material is often non-uniformly strained, and this can also broaden the diffraction lines. On the other hand, the individual crystals, which make up a loose powder of fine particle size, can often be assumed to be strain-free, provided the material involved is a brittle one and all the observed broadening can be ascribed confidently to the crystallite size effect. Then the crystallite sizes could be estimated as approximately 1 nm and 5 nm, respectively. The crystallite of several nanometer-sizes would be similar to that of critical nucleus. Further, it should be noted that these two-profiles have a difference of 1% or less at lattice mismatch, like the super-lattice structure. It is considered that the composition of nucleus would be different from that of γ phase. In contrast, it is considered that the crystallite of 5 nm-size would be the composition of γ phase since the maximum of diffraction profile continuously changes and the X-ray intensity of the main constituent increases with a temperature drop. There is a possibility to get some information of heterogeneous nucleation. According to the theory advanced by Turnbull and Vonnegut, a nucleating agent will be effective in promoting nucleation when the lattice parameters in the low-index crystallographic planes of both the substrate and the nucleated solid are similar. The degree of mismatch of these parameters, i.e., the disregistry, is expressed as $\delta = \Delta a_0/a$. Therefore, it would be very interesting if the crystallite of several nanometer-sizes corresponds to the first nucleus like oxide, carbide and nitride. The disregistry between the catalyst and the matrix was a significant factor for nucleation behaviour since some of the oxides had small disregistries. Though no further information has been analyzed, our technique may give the possibility that understand the behaviour of crystallites in the vicinity of critical-nucleus.

Finally, in the discussion about the FWHM in the vicinity of the melting point, it is necessary to consider the influence of Thermal Diffuse Scattering (TDS). The incoherent part of TDS becomes a background, and the coherent part becomes halo-patterns of liquid phase. It is considered that the former becomes the uniform background and has been deleted by the background removal. The latter is observed about 18 degrees outside this measurement region. It is guessed it might be extremely weak since the second halo-pattern as it guesses from the intensity of the first halo-pattern. Moreover, it is guessed that the thermal vibration might influence the FWHM from the thermal expansion of about 1400°C by about 1.5%. The influence of TDS would be not remarkable.

4. Conclusion

We investigated the directional crystal growth with the information of nucleation by combination of the welding technique and the two-dimensional detector with high time resolution excluding the influence of the crystal preferred orientation. These experimental observations provided *in-situ* diffraction patterns of the phase transformations occurring during rapid weld cooling.

- (1) In Fe-0.88%C, the K-S relationship appeared. It is considered that the original relationship appeared by progress of minimization of microstructure.
- (2) The improvement of crystallinity and the solidification

shrinkage takes after the increase of crystallites size.

- (3) The fraction of shrinkage ratio per Kelvin in the rapid cooling process is $1.9 \times 10^{-5} \text{ K}^{-1}$, which corresponds to the thermal expansion coefficient $2 \times 10^{-5} \text{ K}^{-1}$ of general low carbon steel.
- (4) The maximum tilt is in vicinity of about 650°C, which corresponds to refinements of microstructure.
- (5) There are two tilting zones, one is due to crystallites in the liquid and the other is due to the formation of a microstructure.

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