Numerical Analysis of Hydrogen Diffusion Problems Using the Finite Element Method

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Abstract

This paper focuses on the Krom et al. hydrogen diffusion model in materials. Using the Sofronis and McMeeking model, we reconstruct our own finite element scheme based on the Krom et al. scheme, and apply the Galerkin method in this first trial. Results obtained with our new scheme are compared with previous results obtained by Sofronis and McMeeking, and Krom et al. to show good agreement.

List of symbols

Roman symbols:

$\beta_0[m]$ : Crack tip opening (const).
$b[m]$ : $4.7 \beta_0$ (const).
$C_{L/T}[m^{-3}]$ : Number of hydrogen atoms per unit volume in lattice/trap sites.
$C_b[m^{-3}]$ : Boundary value of $C_L$ (const).
$C_{L0}[m^{-3}]$ : Initial hydrogen concentration (const).
$C_L^*[-]$ : Test function for $C_L$.
$D_L[m^2 \cdot s^{-1}]$ : Diffusion constant (const).

$D^*[-] = \frac{C_L + \theta T}{C_L}$.

$E[N \cdot m^{-2}]$ : Young’s modulus (const).
$f_r(z, \theta)[-]$ : Universal function for plain strain.
$G[N \cdot m^{-2}]$ : Shear modulus (const).
$h[m]$ : Model thickness (const).
$J[s^{-1} \cdot m^{-2}]$ : Hydrogen flux.
$K_i [MPa\sqrt{m}]$ : Stress intensity factor (const).

$K_r [-]$ : Trap equilibrium constant (const).

$M_i [m^2 \cdot s^{-1} \cdot mol \cdot J^{-1}]$ : Mobility of the hydrogen in lattice sites (const).

$n[-]$ : Outward normal vector or time step in the case of superscript.

$n'[-]$ : Hardening exponent (const).

$N_{L,T} [m^{-3}]$ : Number of lattice/trap sites per unit volume ($N_L$; const).

$R [J \cdot mol^{-1} \cdot K^{-1}]$ : Gas constant (const).

$R [m]$ : Model radius (const).

$r [m]$ : Radial component ($r = \sqrt{x^2 + y^2}$).

$T[K]$ : Absolute temperature (const).

$\Gamma [N \cdot m^{-2}]$ : Traction vector.

$t [s]$ : Time.

$u_i [m]$ : Displacement component.

$\tilde{V}_H [m^3 \cdot mol^{-1}]$ : Partial molar volume of hydrogen (const).

$x[m]$ : Coordinate component $x$.

$y[m]$ : Coordinate component $y$.

Greek symbols:

$\Gamma_k$ : On $\Gamma_k$, the number of hydrogen atoms per unit volume is given.

$\Gamma_N$ : On $\Gamma_N$, the flux through the surface is given.

$\partial \Omega = \Gamma$ : Boundary of $\Omega$.

$\Delta E_t [J \cdot mol^{-1}]$ : Trap binding energy (const).

$\Delta t [s]$ : Time interval (const).

$\varepsilon [-]$ : Total strain.

$\varepsilon_p [-]$ : Plastic strain.

$\theta_{L,T} [-]$ : Ratio of the number of occupied sites to the total available for lattice/trap sites.

$\theta [rad]$ : Angle.

$\mu_i [J \cdot mol^{-1}]$ : Chemical potential in lattice sites.

$\mu_i^0 [J \cdot mol^{-1}]$ : Chemical potential in lattice sites at a reference temperature and pressure (const).

$\nu [-]$ : Poisson ratio (const).

$\sigma [N \cdot m^{-2}]$ : Applied stress.

$\sigma_h [N \cdot m^{-2}]$ : Hydrostatic stress.

$\sigma_y [N \cdot m^{-2}]$ : Yield stress (const).

$\sigma_{xy} [N \cdot m^{-2}]$ : Stress tensor component.

$\Omega$ : Domain.
1. Introduction

In spite of decades of research effort, hydrogen assisted cracking is still the most serious problem in the world of hydrogen usage. Hydrogen embrittlement is still not clearly understood, and elastic-plastic stress analysis coupled with diffusion computation is needed to model the effects of hydrogen distribution in a solid. The finite element method (FEM) is a very useful tool for this purpose. P. Sofronis and R. M. McMeeking\(^1\) proposed the FEM model to show the effects of hydrostatic stress and the trapping phenomenon for hydrogen distribution in plastically deformed steel. They investigated the hydrogen concentration near a blunting crack tip under small-scale yielding conditions, but unfortunately, their hydrogen transport model didn't provide the correct balance of hydrogen concentration into the considered material. Based on their work, another model providing the correct balance of hydrogen and showing a strong dependence of the hydrogen concentration in lattice sites on the strain rate was developed by Krom et al.\(^2\).

This paper focuses on this new model. We reconstruct our own finite element scheme and apply the backward time difference with a suitable linearization as the pioneers. The resultant finite element scheme produces a non-symmetric coefficient matrix for each time step, though the original two papers deal with a symmetric matrix for each time step. Our approach is a standard one for the non-linear advection-diffusion equation. However, it has the conservative advection term which depends on absolute values of the hydrostatic stress gradient. So if these absolute values are larger, then the Péclet number becomes larger and some numerical oscillations will appear. One remedy is a well known stabilization technique. However the stabilization technique for the conservative advection term is not well known. So, the Galerkin method is applied in this first trial. In fact, some numerical difficulties are encountered for great absolute values of the hydrostatic stress gradient. This problem will be overcome by the use of the stabilization technique, for example. With a fully coupled analysis, they are our next goals for hydrogen diffusion problems in materials, and will be mentioned in the forthcoming papers.

2. The hydrogen transport equation

2.1 The original model

![Diagram](image)

\(C_t (R, \theta, t) = C_s \text{ on } \Gamma_g\)

\(C_t (R, \theta; 0) = C_{L0} \text{ in } \Omega\)

\(C_s = C_b \text{ on } \Gamma_g\)

\(T = 0\)

\(T_s = 0 \text{ on } \Gamma_N = \partial \Omega - \Gamma_g\)

\(J(r, 0, t) \cdot n = 0 \quad u_f = 0\)

\(t = \text{time}\)

Fig 1. Description of the boundary and initial conditions for the one side coupled diffusion and elastic-plastic problem under small scale yielding conditions\(^1\).
The original model is based on the Sofronis and McMeeking model \(^1\), which is based on the equilibrium theory established by Oriani \(^3\). Hydrogen transport between trap sites is made by lattice diffusion \(^2\). Trap sites can be found at micro structural defects \(^14\), and are assumed not to form an extended network; they are supposed to be reversible and saturable \(^11\), \(^2\). The main difference between the Sofronis and McMeeking model and the Krom et al. model \(^2\) is a factor depending on the strain rate in order to get the correct balance of hydrogen concentration.

In Fig 1, we consider a body with the volume \(\Omega\) and the surface \(\partial\Omega = \Gamma\). Because of the mass conservation law, the rate of change of total hydrogen inside \(\Omega\) is equivalent to the flux through \(\Gamma\). Constants \(C_h\) and \(C_{j,0}\) are given numbers of hydrogen atoms per unit volume in Normal Interstitial Lattice Sites (NILS). \(J\) is the hydrogen flux, \(u\) displacement components, and \(T\) the traction vector \(^1\).

\[
\begin{align*}
\frac{\partial}{\partial t} \int_{\Omega} \left[ (C_i + C_j) \right] d\Omega + \int_{\partial \Omega} J \cdot ndS &= 0, \\
J &= -M_j C_j \nabla \mu_i, \quad \mu_i = \mu_i^0 + RT \ln \frac{C_j}{N_j} - \overline{V}_m \sigma_h.
\end{align*}
\]

(1) \(C_i\) and \(C_j\) are respectively the number of hydrogen atoms per unit volume in lattice sites and in trap sites. \(J\) is the hydrogen flux, \(n\) is the outward normal vector, \(M_j\) is the mobility of the hydrogen in the lattice sites, \(N_j\) is the number of lattice sites per unit volume, and \(\mu_i\) is the chemical potential in lattice sites. \(R\) is the gas constant, \(T\) is the absolute temperature, \(\mu_i^0\) is the chemical potential at a reference temperature and pressure, and \(\sigma_h\) is defined as the hydrostatic stress, such that \(\sigma_h = \frac{1}{3} \sum_{i=1}^{3} \sigma_{ii}\), where the component \(\sigma_{ii}\) is the stress tensor component. The subscripts \(i\) and \(j\) are usually denoted \(x, y, z\) instead of \(1, 2, 3\). \(\overline{V}_m\) is the partial molar volume of hydrogen.

Using the expression of \(\mu_i\), the hydrogen flux can be expressed with the lattice diffusivity \(D_j\) as

\[
J = -D_j \nabla C_j + \frac{D_j C_j \overline{V}_m}{RT} \nabla \sigma_h.
\]

(2) \(D_j = M_j RT\).

After substituting (2) in (1), and applying the divergence theorem, we obtain for the arbitrary volume \(\Omega\)

\[
\frac{\partial C_i}{\partial t} + \frac{\partial C_j}{\partial t} - \nabla \cdot (D_j \nabla C_j) = \nabla \cdot \left( \frac{C_j D_j \overline{V}_m}{RT} \nabla \sigma_h \right) = 0.
\]

(3)

The hydrogen concentrations are first related to the numbers of sites

\[
C_i = \theta_i N_i, \quad \text{ (4)}
\]

\[
C_j = \theta_j N_j, \quad \text{ (5)}
\]
where $\theta_c$ and $\theta_f$ are respectively the occupancy of lattice sites and the occupancy of trap sites, and $N_c$ and $N_f$ are the lattice site and the trap site densities. An expression of the hydrogen concentration in trap sites as a function of concentration in lattice sites given by Oriani [3] in the case of the equilibrium is, under the assumption that $\theta_f << 1$,

$$C_f = \frac{N_f}{1 + \frac{1}{K_f \theta_f}}$$

where $K_f$ is the equilibrium constant such that $K_f = e^{-N_f/(RT)}$ with $\Delta E_f$ as the trap binding energy. The partial derivative of the hydrogen concentration in trap sites with respect to time is

$$\frac{\partial C_f}{\partial t} = \frac{\partial C_f}{\partial C_L} \frac{\partial C_L}{\partial t} + \frac{\partial C_f}{\partial N_f} \frac{dN_f}{dt} - \frac{\partial C_f}{\partial \epsilon_p} \frac{d\epsilon_p}{dt}$$

with $\epsilon_p$ as the plastic strain. The final equation presented by Krom et al. [3] for hydrogen transport is

$$\frac{C_f}{C_f} + C_f (1 - \theta_f) \frac{\partial C_f}{\partial t} - \nabla \cdot (D_f \nabla C_f) + \nabla \left( \frac{C_f D_f \bar{V}_H}{RT} \nabla \sigma_h \right) + \theta_f \frac{dN_f}{dt} \frac{\partial \epsilon_p}{\partial t} = 0.$$  

2.2 The weak form

The multiplication by a test function of (8) and integration over $\Omega$ produces

$$\int_{\Omega'} \left( D^* \frac{\partial C_L}{\partial t} - \nabla \cdot (D_f \nabla C_f) + \nabla \left( \frac{C_f D_f \bar{V}_H}{RT} \nabla \sigma_h \right) + \theta_f \frac{dN_f}{dt} \frac{\partial \epsilon_p}{\partial t} \right) d\Omega = 0.\tag{9}$$

where $C_f^*$ is the test function, and using the Green's theorem in (9), the following result is obtained.

$$\int_{\partial \Omega} \left( C_f^* D^* \frac{\partial C_f}{\partial t} + \nabla C_f^* \cdot D_f \nabla C_f - \nabla C_f^* \cdot \frac{C_f D_f \bar{V}_H}{RT} \nabla \sigma_h \right) + \theta_f \frac{dN_f}{dt} \frac{\partial \epsilon_p}{\partial t} \right) \cdot n dS = 0.\tag{10}$$

where

$$D^* = \frac{C_f + C_f (1 - \theta_f)}{C_f}.$$  

$$\int_{\partial \Omega} \left( C_f^* \cdot n dS = 0.\tag{11}$$
Considering the concentration fixed on the surface $\Gamma_\epsilon$, we have $C_\epsilon^* = 0$ on $\Gamma_\epsilon$. Also, we have $J \cdot n = 0$ on $\Gamma_N$. Therefore we have $\int_{\Gamma_N} C_\epsilon^* J \cdot n \, dS = 0$. A detailed expression of $\theta_\epsilon$ as a function of $C_\epsilon$ is needed as follows.

Using (4), (5) and (6), we can find

$$\theta_\epsilon = \frac{1}{1 + \frac{N_\epsilon}{K_\epsilon C_\epsilon}} = \frac{K_\epsilon C_\epsilon}{K_\epsilon C_\epsilon + N_\epsilon}.$$  \hspace{1cm} (12)

Also, we have

$$C_\epsilon = \frac{N_\epsilon K_\epsilon}{K_\epsilon + \frac{N_\epsilon}{C_\epsilon}} = \frac{C_\epsilon N_\epsilon K_\epsilon}{K_\epsilon C_\epsilon + N_\epsilon}.$$  \hspace{1cm} (13)

Then, after using (12) and (13), $D^*$ has the following expression

$$D^* = 1 + \left[ \frac{N_\epsilon K_\epsilon N_\epsilon}{(K_\epsilon C_\epsilon + N_\epsilon)^2} \right].$$  \hspace{1cm} (14)

Having the detailed expressions of $\theta_\epsilon$ in (12) and $D^*$ in (14), we finally obtain the following equation:

$$\left( 1 + \frac{N_\epsilon K_\epsilon N_\epsilon}{(K_\epsilon C_\epsilon + N_\epsilon)^2} \right) \partial_t C_\epsilon^* + \left( D_\epsilon \nabla C_\epsilon, \nabla C_\epsilon^* \right)$$

$$- \left( \frac{C_\epsilon D_\epsilon^*_\epsilon}{RT} \nabla \sigma_a, \nabla C_\epsilon^* \right) + \left( C_\epsilon \frac{K_\epsilon}{K_\epsilon C_\epsilon + N_\epsilon} \frac{dN_\epsilon}{dt} \frac{\partial \xi^*}{\partial \xi}, C_\epsilon^* \right) = 0,$$  \hspace{1cm} (15)

where $(A, B) = \int_{\Omega} A \cdot B d\Omega$.

3. **Our numerical scheme**

For our numerical scheme, we make the choice that all the terms have unknown parts. The motivation of this choice is to avoid some computational difficulties which appear with the increase of $|\nabla \sigma_a|$ in the Krom et al. scheme $^2$. By this new scheme, we can solve high Péclet number problems with suitable stabilization techniques.

Furthermore, we need an explicit form for $\frac{dN_\epsilon}{dt}$ and $\frac{\partial \xi^*}{\partial \xi}$. For the number of trap sites $N_\tau$, the following expression, also used by Sofronis and McMeeking$^{1}$ and Krom et al.$^2$ before, is used:
\[
\log_{10}(N_j) = 23.26 - 2.33 e^{-5.5\varepsilon_j}.
\]

(16)

Using this relation, we can have the expression for \(\frac{dN_j}{d\varepsilon_j} \cdot \frac{\partial \varepsilon_j}{\partial t}\):

\[
\begin{align*}
\frac{\partial \varepsilon_j}{\partial t} & \approx \frac{\varepsilon_j^{n+1} - \varepsilon_j^n}{\Delta t}, \\
\frac{dN_j}{d\varepsilon_j} & = N_j \cdot \log_{10}(10 \cdot 12.815) e^{-5.5\varepsilon_j}.
\end{align*}
\]

(17)

After having the expression for \(\frac{dN_j \cdot \partial \varepsilon_j}{d\varepsilon_j} \frac{\partial \varepsilon_j}{\partial t}\) with the finite difference in time (\(\Delta t\): the time step) in (17), we get the following implicit scheme,

\[
\left[ \left( 1 + \frac{N_j K_j N_j (\varepsilon_j^n)}{K_j C_j^n + N_j} \right) \frac{C_j^{n+1} - C_j^n}{\Delta t}, C_j^n \right] + \left( D_j \nabla C_j^{n+1}, \nabla C_j^n \right) - \left( \frac{C_j^{n+1} D_j \nabla \sigma_j^n \cdot \nabla C_j^n}{RT} \right) + \left( \frac{C_j^{n+1} 29.508 K_j N_j (\varepsilon_j^n) e^{-5.5\varepsilon_j}}{K_j C_j^n + N_j} \frac{\varepsilon_j^{n+1} - \varepsilon_j^n}{\Delta t}, C_j^n \right) = 0.
\]

(18)

As we can see, the above scheme is different from the following Krom et al. implicit scheme,

\[
\left[ \left( 1 + \frac{N_j K_j N_j (\varepsilon_j^n)}{K_j C_j^n + N_j} \right) \frac{C_j^{n+1} - C_j^n}{\Delta t}, C_j^n \right] + \left( D_j \nabla C_j^{n+1}, \nabla C_j^n \right) =
\]

\[
\left( \frac{C_j^n D_j \nabla \sigma_j^n \cdot \nabla C_j^n}{RT} \right) + \left( \frac{C_j^{n+1} 29.508 K_j N_j (\varepsilon_j^n) e^{-5.5\varepsilon_j}}{K_j C_j^n + N_j} \frac{\varepsilon_j^{n+1} - \varepsilon_j^n}{\Delta t}, C_j^n \right).
\]

(19)

It is noted that (18) produces a non-symmetric coefficient matrix for each time step due to the third term in the left hand side, though (19) makes a symmetric coefficient matrix. It is also noted that \(\varepsilon_j^{n+1}\) in (18) and (19) is given with \(\varepsilon_j^n\) because a structural analysis is performed before the diffusion analysis of hydrogen concentration. The hydrostatic stress \(\sigma_j^n\) is also provided by the stress analysis in the same way. Our final remark is that the backward time difference is adopted as the pioneers\(^{1,2}\) because of its excellent stability character in the sense of the maximum norm.
4. Numerical data and results

4.1 A mesh, boundary conditions and material properties

Some other differences between the Krom et al. computation and ours are mentioned. The software used to perform the stress analysis in our case is ADVENTURE_Solid and related modules; we work on a three-dimensional model, so we use finite elements different from those used by Krom et al.:

- Krom et al.: Q1 elements (2D case),
- Our computation: P2 elements (3D case), see Fig 2 and Fig 3.

The plastic strain and hydrostatic stress data are also obtained before the hydrogen diffusion analysis by using the ADVENTURE Solid software.

![Fig 2. A mesh.](image)

<table>
<thead>
<tr>
<th>Table 1. Model dimensions.</th>
</tr>
</thead>
<tbody>
<tr>
<td>Crack tip opening $b_0$</td>
</tr>
<tr>
<td>Model radius $R$</td>
</tr>
<tr>
<td>Model thickness $h$</td>
</tr>
</tbody>
</table>

The model is a half circle with a radius of $1.5 \times 10^{-1}$ [m] and a thickness of $8.25 \times 10^{-4}$ [m]. The crack can be considered as a horizontal band, going from the center to the left until the circumference of the half circle with a thickness of $1.0 \times 10^{-3}$ [m]. A recap of the model dimension is found in Table 1.

The mesh consists of 10,656 nodes and 3,706 10-node tetrahedral elements. The mesh is refined near the crack tip.

![Fig 3. The mesh near the crack tip.](image)
Numerical Analysis of Hydrogen Diffusion Problems

Table 2. Numerical parameters.

<table>
<thead>
<tr>
<th>Parameter</th>
<th>Value</th>
</tr>
</thead>
<tbody>
<tr>
<td>Number of elements</td>
<td>3,706</td>
</tr>
<tr>
<td>Number of nodes</td>
<td>10,656</td>
</tr>
<tr>
<td>CPU</td>
<td>Pentium 4 3.2 GHz</td>
</tr>
<tr>
<td>Memory</td>
<td>2.0 GB</td>
</tr>
<tr>
<td>CPU time</td>
<td>480–600[s]</td>
</tr>
<tr>
<td>Time increment $\Delta t$</td>
<td>0.005[s], 0.5 [s], etc…</td>
</tr>
<tr>
<td>Total time</td>
<td>1.3[s], 130[s], etc…</td>
</tr>
<tr>
<td>Number of total steps</td>
<td>260</td>
</tr>
<tr>
<td>Solver</td>
<td>BiCGSTAB(L): L=4</td>
</tr>
<tr>
<td>Preconditioner</td>
<td>Shifted ILU (shift value: 1.05)</td>
</tr>
<tr>
<td>Convergence criterion for relative residuals in BiCGSTAB (L.)</td>
<td>$10^{-6}$</td>
</tr>
</tbody>
</table>

The BiCGSTAB(L) solver \(^{41}\) is used with L=4 for the non-symmetric coefficient matrix. The ILU preconditioner is also used with the acceleration parameter which is set to be 1.05. **Table 2** recapitulates the set of numerical parameters used in our computation.

The diffusion analysis and the stress analysis were carried separately. Results produced by the stress analysis such as the hydrostatic stress and the strain rate factor were used to perform the diffusion analysis. Boundary conditions are given as in **Fig. 1**. Normally, the boundary condition $C_h$ is given on $\Gamma_h$ and set to be $C_{L,0}$ in this computation. On $\Gamma_n = \partial \Omega - \Gamma_h$ (The symmetry line $\theta = 0$), the hydrogen flux is given as $J \cdot n = 0$. For stress analysis, $u_y = 0$ and $T_y = 0$ on $\Gamma_n$, $T = 0$ on the surface of the crack, and the displacement $u_x$ (i=x or y) is given on the other boundary as asymptotic boundary conditions of the singular linear elastic field for the mode I opening ($K_f = 89.2 MPa\sqrt{m}$). Here, boundary conditions are explained as for the two dimensional case, but the three dimensional case is straightforward. The diffusion coefficient $D_h$ is $1.27 \times 10^{-8} [m^2 \cdot s^{-1}]$ and the initial hydrogen concentration $C_{L,0}$ is $2.08 \times 10^{21} [m^{-3}]$ \(^{121}\) ($4.425 \times 10^{4}$ [ppm]). Stress data are provided for 260 steps with a time increment of 0.5[s], which means a total time of 130[s]. The temperature $T$ is fixed at $300[K]$ and the value of the partial molar volume $V_m$ is $2.0 \times 10^{-5} [m^3 \cdot mol^{-1}]$. The number of lattice sites per unit volume $N_f$ is $5.10 \times 10^{28} [m^{-3}]$ and the trap binding energy $\Delta E_T$ is $-60 \times 10^{-5}$ $[J \cdot mol^{-1}]$. **Table 3** is a recap of the different constant values.

We use the same stress-strain relation as the one used by Krom et al. \(^2\),

$$
\varepsilon = \begin{cases} 
\frac{\sigma}{E} & \text{if } \varepsilon \leq \frac{\sigma_s}{E} \\
\frac{\sigma}{E} \left( \frac{\sigma}{\sigma_s} \right)^\nu & \text{if } \varepsilon > \frac{\sigma_s}{E} 
\end{cases}
$$

(20)
where $E$ is the Young's modulus: 207[MPa], $\sigma[N \cdot m^{-2}]$ is the applied stress, $\varepsilon[-]$ is the total strain, $\sigma_y = 250[MPa]$ is the yield stress, and $n = 5$ is the hardening exponent. To be able to use this power law relation, a modification of the ADVENTURE_Solid software has been necessary. This relation is given by Cauchy stresses, but the relation used by Sofronis and McMeeking is different, because it is given by Kirchoff stresses.

**Table 3. Constant values.**

<table>
<thead>
<tr>
<th></th>
<th>$D_L$</th>
<th>$1.27 \times 10^{-8}[m^2 \cdot s^{-1}]$</th>
</tr>
</thead>
<tbody>
<tr>
<td>Gas constant</td>
<td>$R$</td>
<td>$8.3144[J \cdot mol^{-1} \cdot K^{-1}]$</td>
</tr>
<tr>
<td>Temperature</td>
<td>$T$</td>
<td>$300[K]$</td>
</tr>
<tr>
<td>Initial hydrogen concentration</td>
<td>$C_{i,0}$</td>
<td>$2.08 \times 10^{11}[m^{-3}]$</td>
</tr>
<tr>
<td>Partial molar volume of hydrogen</td>
<td>$\bar{V}_{ii}$</td>
<td>$2 \times 10^{-5}[m^3 \cdot mol^{-1}]$</td>
</tr>
<tr>
<td>Lattice sites per unit volume</td>
<td>$N_{i}$</td>
<td>$5.10 \times 10^{25}[m^{-3}]$</td>
</tr>
<tr>
<td>Trap binding energy</td>
<td>$\Delta E_f$</td>
<td>$-60 \times 10^{-3}[J \cdot mol^{-1}]$</td>
</tr>
<tr>
<td>Poisson ratio</td>
<td>$\nu$</td>
<td>$0.3[-]$</td>
</tr>
<tr>
<td>Young modulus</td>
<td>$E$</td>
<td>207[MPa]</td>
</tr>
<tr>
<td>Yield stress</td>
<td>$\sigma_y$</td>
<td>$250[MPa]$</td>
</tr>
</tbody>
</table>

**4.2 Results**

We first consider stress analysis results, comparing our hydrostatic stress (the vertical axis) distribution ahead of the crack tip in the line ($\theta = 0^\circ$) with those of two previous papers in Fig 4. We have on the horizontal axis the ratio on the radial component $r$ over the crack tip opening displacement $b = 4.7b_0$. The crack grows at each step, the stress intensity factor is $K_f = 89.2[MPa\sqrt{m}]$, $\sigma_h$ is the hydrostatic stress, and $\sigma_y$ is the yield stress. The results are provided for a loading time of 130[s].

As we can see in Fig 4, our results match well with Krom et al. hydrostatic stress. But some differences are observed around the center of the model. This is probably due to the difference in the type of elements, especially around the center where our mesh is refined.

Now we compare the hydrogen distribution in lattice sites at 130[s] with Krom et al. results in Fig 5. The results with the strain rate factor and without the strain rate factor are computed for a loading time of 130[s]. $C_{i,0}$ is the initial hydrogen concentration, and again $\theta = 0^\circ$. Effects of the strain rate factor can be observed in both results. Good agreement is shown.
Fig 4. Hydrostatic stress comparison.

\[ \text{Fig 5: } \frac{C_L}{C_{L0}} \text{ ahead of the crack tip by our scheme and by Krom et al. scheme at } 130[s] \text{ (end of loading).} \]

We extend the comparison of hydrogen distribution in lattice sites by computing it for several loading times. We keep \( K_f = 89.2 [\text{MPa} \sqrt{\text{m}}] \), and the hydrogen concentration is prescribed on the crack surface.

We notice that the hydrogen concentration obtained with our scheme is the same as the concentration obtained with the Krom et al. scheme for several loading times in Fig 6.

In Fig 7, we observe the variations of hydrogen concentration in trap sites. In Krom et al. results, hydrogen concentration distribution in trap sites is given for a loading time of 1.3[s], and for loading times going from 3.25[s] to \( 1.3 \times 10^6[s] \). The concentration of hydrogen in trap sites is sensibly the same. As we can see, our results are quite close to Krom et al. results \(^{3}\). In our case, we also consider a loading time of 1.3[s], and we consider another loading time included between 3.25[s] and \( 1.3 \times 10^6[s] \), i.e., 130[s]. The concentration peak in trap sites is approximately 86 times the initial concentration in lattice sites. According to Krom et al., even though the strain has a slight effect on hydrogen concentration in trap sites for the considered loading times, the trapping of hydrogen is necessary to get a significant impact of the strain rate on hydrogen concentration in lattice sites.
5. Conclusion

Using the Sofronis and McMeeking model, we have reconstructed our own finite element scheme based on the Krom et al. scheme, and applied the Galerkin method in this first trial. Results obtained with our new scheme have shown that the scheme can provide good results for this kind problem. Some numerical complications may be encountered for great values of $|\nabla \sigma_h|$. They could be improved by the use of the stabilized method, for example.

Fully coupled analysis may also be required; in order to perform a fully coupled analysis, the hydrostatic stress and the strain rate factor should fully depend on the total hydrogen concentration in lattice and trap sites \(^7,8\). Those are our next goals for hydrogen diffusion problems in materials.

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